



CHEMISTRY

CHEAT SHEET



NAMING SIMPLE COMPOUNDS

NAMING SIMPLE COMPOUNDS CHEAT SHEET

Naming Compounds with Their Formulas
(Show the ions involved in the compound)



Steps for Naming:

1. Write the formula of the compound.
2. Identify the cation and anion.
3. Name the cation first, then the anion.
4. If the cation is a transition metal, include its oxidation state in parentheses.
5. If the anion is a polyatomic ion, use its name.
6. If the compound is a binary ionic compound, use the element names.

Binary Compounds		Common Names	
Formula	Name	Formula	Name
NaCl	Sodium Chloride	NaCl	Salt
KCl	Potassium Chloride	KCl	Salt
CaCl ₂	Calcium Chloride	CaCl ₂	Salt
MgCl ₂	Magnesium Chloride	MgCl ₂	Salt
AlCl ₃	Aluminum Chloride	AlCl ₃	Salt
FeCl ₂	Iron(II) Chloride	FeCl ₂	Salt
FeCl ₃	Iron(III) Chloride	FeCl ₃	Salt
CuCl ₂	Copper(II) Chloride	CuCl ₂	Salt
ZnCl ₂	Zinc Chloride	ZnCl ₂	Salt
AgCl	Silver Chloride	AgCl	Salt

Naming Compounds with a Central and Peripheral



Steps for Naming:

1. Write the formula of the compound.
2. Identify the central atom and the peripheral atoms.
3. Name the central atom first, then the peripheral atoms.
4. If the peripheral atoms are polyatomic ions, use their names.

Common Names

Formula	Central Atom	Peripheral Atoms	Common Name
NaCl	Na	Cl	Salt
KCl	K	Cl	Salt
CaCl ₂	Ca	Cl	Salt
MgCl ₂	Mg	Cl	Salt
AlCl ₃	Al	Cl	Salt
FeCl ₂	Fe	Cl	Salt
FeCl ₃	Fe	Cl	Salt
CuCl ₂	Cu	Cl	Salt
ZnCl ₂	Zn	Cl	Salt
AgCl	Ag	Cl	Salt

Common Names for Specific Compounds

Formula	Central Atom	Peripheral Atoms	Common Name
NaCl	Na	Cl	Salt
KCl	K	Cl	Salt
CaCl ₂	Ca	Cl	Salt
MgCl ₂	Mg	Cl	Salt
AlCl ₃	Al	Cl	Salt

Naming Compounds Study Guide

Kathleen Armour



Naming Compounds Study Guide:

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