

### Common Polyatomic Ions

(Alphabetical order by ion name)

**NOTE:** -ite ending means one less oxygen than the -ate form.

Ion Name	Ion Symbol	Ion Name	Ion Symbol	Ion Name	Ion Symbol	Ion Name	Ion Symbol
Acetate	$\text{CH}_3\text{CO}_2^-$ or $\text{CH}_3\text{COO}^-$	Dichromate	$\text{Cr}_2\text{O}_7^{2-}$	Hypobromite	$\text{BrO}^-$	Perrhenate	$\text{ReO}_4^-$
Ammonium	$\text{NH}_4^+$	Dihydrogen Phosphate	$\text{H}_2\text{PO}_4^-$	Hypochlorite	$\text{ClO}^-$	Phosphate	$\text{PO}_4^{3-}$
Arsenate	$\text{AsO}_4^{3-}$	Dihydrogen Phosphite	$\text{H}_2\text{PO}_3^-$	Hypoiodite	$\text{IO}^-$	Phosphite	$\text{PO}_3^{3-}$
Arsenite	$\text{AsO}_3^{3-}$	Dithionate	$\text{S}_2\text{O}_6^{2-}$	Iodate	$\text{IO}_3^-$	Plumbate	$\text{PbO}_3^{2-}$
Azide	$\text{N}_3^-$	Dithionite	$\text{S}_2\text{O}_4^{2-}$	Iodite	$\text{IO}_2^-$	Plumbite	$\text{PbO}_2^{2-}$
Borate	$\text{BO}_3^{2-}$	Ferricyanide	$\text{Fe}(\text{CN})_6^{3-}$	Isocyanate	$\text{NCO}^-$	Rhenate	$\text{ReO}_4^{2-}$
Bromate	$\text{BrO}_3^-$	Ferrocyanide	$\text{Fe}(\text{CN})_6^{4-}$	Manganate	$\text{MnO}_4^{2-}$	Selenate	$\text{SeO}_4^{2-}$
Bromite	$\text{BrO}_2^-$	Fulminate	$\text{CNO}^-$	Nitrate	$\text{NO}_3^-$	Selenite	$\text{SeO}_3^{2-}$
Carbonate	$\text{CO}_3^{2-}$	Hydrazide	$\text{N}_2\text{H}_3^-$	Nitrite	$\text{NO}_2^-$	Silicate	$\text{SiO}_3^{2-}$
Chlorate	$\text{ClO}_3^-$	Hydrogen arsenate	$\text{HAsO}_4^{2-}$	Oxalate	$\text{C}_2\text{O}_4^{2-}$	Stannate	$\text{SnO}_3^{2-}$
Chlorite	$\text{ClO}_2^-$	Hydrogen carbonate	$\text{HCO}_3^-$	Ozonide	$\text{O}_3^-$	Stannite	$\text{SnO}_2^{2-}$
Chromate	$\text{CrO}_4^{2-}$	Hydrogen Phosphate	$\text{HPO}_4^{2-}$	Perbromate	$\text{BrO}_4^-$	Sulfate	$\text{SO}_4^{2-}$
Chromite	$\text{CrO}_2^-$	Hydrogen Phosphite	$\text{HPO}_3^{2-}$	Perchlorate	$\text{ClO}_4^-$	Sulfite	$\text{SO}_3^{2-}$
Citrate	$\text{C}_6\text{H}_5\text{O}_7^{3-}$	Hydrogen Sulfate	$\text{HSO}_4^-$	Periodate	$\text{IO}_4^-$	Superoxide	$\text{O}_2^-$
Cyanate	$\text{OCN}^-$	Hydrogen Sulfite	$\text{HSO}_3^-$	Permanganate	$\text{MnO}_4^-$	Thiocyanate	$\text{SCN}^-$
Cyanide	$\text{CN}^-$	Hydroxide	$\text{OH}^-$	Peroxide	$\text{O}_2^{2-}$	Thiosulfate	$\text{S}_2\text{O}_3^{2-}$
						Tungstate	$\text{WO}_4^{2-}$

#### PREFIXES:

per- = one more oxygen than -ate

hypo- = one less oxygen than -ite

# Naming Polyatomic Ions Packet Answers Pogill

**AW Chickering**



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