

Biology 1010 Midterm Study Guide

The Biology 1010 midterm will cover material specifically from Modules 2-6. The exam will consist of multiple-choice, true/false, matching types of questions. You will have an hour to complete around 45-50 questions. (The vast majority of students complete the exam well under the time limit.) This study guide is provided for your reference and study only. It is not an assignment. You may create a single index card (4x6" size max) of notes to use on the exam. This is intended to be a hand-written card. It can include writing front and back. Those are the only notes allowed. You may not use any other notes, textbooks, phone, online resources, etc. Practice questions are a great way to study so use your Weekly Quizzes as review materials.

Module 2:

- Be familiar with the three types of bonds discussed in class and what is happening with electrons in each case:
 - **ionic – electrons are given and taken; ions are formed
 - **covalent – characterized by shared electrons
 - **know the difference between nonpolar and polar covalent bonds
 - **hydrogen – form between partial charges on H-containing molecules; for example, between the negative oxygen end of one water molecule and the positive hydrogen end of another water molecule
- Recognize that water is polar. A single water molecule consists of two polar covalent bonds between hydrogens and oxygen resulting in polarity.
- Because of water's polarity and the hydrogen bonds that form between water molecules - water is cohesive, ice floats, and water is a good solvent.
- Carbon is the backbone of all organic molecules
- Macromolecule Review

Carbohydrates Elements: C, H, O Monomer: monosaccharide ex. glucose Polymer Examples: <ul style="list-style-type: none"> • starch – • cellulose – • glycogen - 	Lipids Elements: C, H, O Components of a fat: <ol style="list-style-type: none"> 1. glycerol 2. fatty acid(s) Defining characteristic of lipids: hydrophobic (non-polar)
Proteins Elements: C, H, O, N Monomer: amino acids A polypeptide is a chain of amino acids. Protein shape determines function.	Nucleic Acids Elements: C, H, O, N, P Monomer: nucleic acids Polymer examples: DNA and RNA

Module 3:

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Wenbin Ji



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