

Microsoft Access Help Manual

Daniela Niemeyer

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web use the link phet colorado en simulation isotopes and atomic mass to go to the phet website for the simulation isotopes and atomic mass begin with the isotopes and atomic mass click on the symbols next to the bar with symbol and abundance in nature to display the data

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web multiple choice 1 minute 1 pt lithium has only two stable isotopes use the sim to determine the average atomic mass of a sample containing three lithium 6 atoms and two lithium 7 atoms 6 51556 amu 6 34875 amu 6 41548 amu

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web write at least three conclusions from your exploration 1 protons and neutrons determine the mass of the atom 2 only protons determine the atomic number 3 the more unstable the atom the less presence or no presence in nature 1 protons and neutrons determine the mass of the atom 2 only protons determine the atomic number 3 w z d v u d p v Á z z v u d x - Dec 27 2021

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web this guided inquiry activity has two learning goals students will be able to students will be able to 1 explain the difference between atomic mass and mass number and 2 calculate average atomic mass from percent abundance and isotopic mass

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web find the average atomic mass of an element given the abundance and mass of its isotopes predict how the mass and name of an isotope will change given a change in the number of protons neutrons or electrons

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