

Limiting Reagent and Percentage Yield

- 1 Dichlorine monoxide is produced by passing chlorine gas over heated mercury (II) oxide according to the following unbalanced reaction:



Balance the reaction. Suppose the quantity of the reactants is sufficient to produce 0.86 g of Cl_2O , but only 0.71 g is obtained. What is the percentage yield?

- 2 Consider the following reaction:



The theoretical yield of beryllium chloride was 10.7 g. Suppose the reaction actually yields 4.5 g. What is the percentage yield?

- 3 Nitrogen and hydrogen at high temperatures are converted to ammonia using the following reaction:



When 400 g of H_2 are added to an excess amount of N_2 , 104 g of NH_3 are formed. Calculate the percentage yield.

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John T. Moore, Richard H. Langley



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Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on

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Experimental Organic Chemistry Clark F. Most,1988-02-09 Acquaints students with all basic laboratory procedures coordinating enough theory and technique to enable readers to fully comprehend the reactions being studied and the procedures involved Material is organized in four sections techniques experiments organic qualitative analysis and appendixes The first section introduces students to all common organic techniques and provides an illustrative experiment with each A unique format helps train the research oriented student to look for relationships that are not immediately apparent The experiments section moves on to more complex experiments involving synthetic procedures followed by work up and analysis requiring more than one technique Instructions are complete and easy to follow and a set of pre laboratory experiments encourages students to determine goals before beginning lab work The appendixes cover less referred to techniques sublimation density determination and molecular weight determinations and contain a pronunciation guide and a compilation of chemical hazards

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