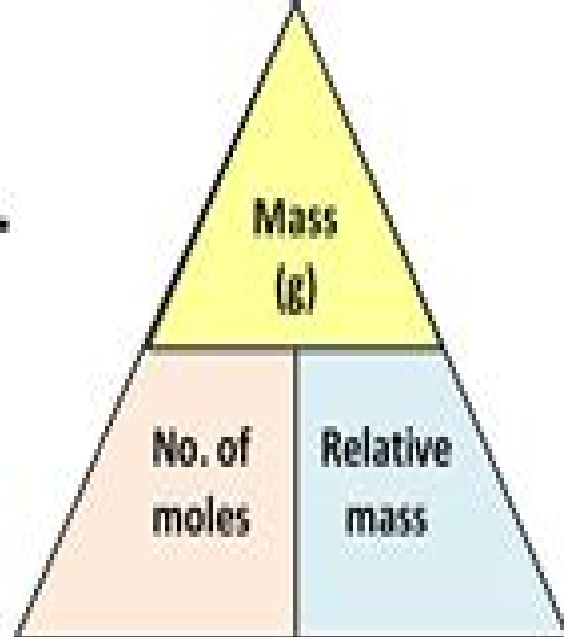


Calculating limiting reactants.

Let's look at an example:



In this reaction, 56 g nitrogen gas (N_2) was added to a chamber containing 14 g hydrogen gas (H_2). Which was the limiting reactant? (A_r : $\text{H}=1$, $\text{N}=14$)

First steps?

Limiting Reactants C1y Vm

WJ Hussar

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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