Limiting Reactant vs Excess Reactant

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Limiting Reactant

Excess Reactant

Limiting reactant is the reactant that can limit the formation of the final product in a chemical reaction

Excess reactant is the reactant that is present in excess in a reaction mixture

DESCRIPTION

DEFINITION

Completely consumed during the chemical reaction Not all reactant molecules are consumed in the reaction

END OF THE REACTION

No limiting reactant present Excess reactant is present

PRODUCT

Limits the amount of final product formed No effect on the final product

Limiting Reactants Pogil Key

IM Harris

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