# Limiting Reactant, Theoretical Yield, Actual Yield, & Percent Yield 22,99 + 35,45 = 58,44

If we have 53.2 g of Na and 65.8 g of Cl<sub>2</sub>, what are the limiting reactant and theoretical yield?

# **Limiting Reagent And Percent Yield 9 3 Answers**

**Richard Bailey** 

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