Read the given statements and select the correct option.

Statement 1: The formula connecting u,v and

f for a spherical mirror is valid only for mirrors whose sizes are very small compared to their radii of curvature.

Statement 2: Laws of reflection are strictly valid for plane surfaces, but not for large spherical surfaces.

- (A) Both statements 1 and 2 are true and statement 2 is the correct explanation of statement 1.
   (B) Both statements 1 and 2 are true but
- statement 2 is not the correct explanation of statement 1. (C) Statement 1 is true but statement 2 is false.
- (D) Statement 1 is false but statement 2 is true.
- 2. A boy throws a table tennis ball of mass 20 g upwards with a velocity of u<sub>0</sub> = 10 m/s at an angle θ<sub>0</sub> with the vertical. The wind imparts a horizontal force of 0.08 N, so that the ball returns to the starting point. Then, the angle θ<sub>0</sub> must be such that, tan θ<sub>0</sub> is

(B) 0.4

(A) 0.2

 A weight is attached to the free end of a sonometer wire. It gives resonance at a length 40 cm when it is resonanced with a tuning fork of frequency 51 Hz. The weight is then immersed

(C) 2.5

(D) 1.2

- wholly in water, the resonant length is reduced to 30 cm. The relative density in which weight suspended is (A) 16/9 (B) 16/7 (C) 16/5 (D) 16/3
- 4. Hydrogen sulphide (H<sub>2</sub>S) contains 94.11% sulphur, water (H<sub>2</sub>O) contains 11.11% hydrogen and sulphur dioxide (SO<sub>2</sub>) contains 50% oxygen. Find the ratio of all given elements. After your calculations which law has been verified?
  (A) Law of multiple proportions
  - (B) Law of reciprocal proportions
     (C) Law of constant composition
     (D) Law of combining volumes
- 5. Which of the following represents the given mode of hybridisation, sp<sup>2</sup>-sp<sup>2</sup>-sp-sp from left to right?

(A) CH<sub>3</sub>=CH−C≡N (B) CH≡C−C≡CH

- (C) CH2=C=C=CH2 (D)
- 6. In the reaction, 4NH<sub>3(g)</sub> + 5O<sub>2(g)</sub> → 4NO<sub>(g)</sub> + 6H<sub>2</sub>O<sub>(f)</sub> when 1 mole of ammonia and 1 mole of O<sub>2</sub> are made to react to completion
  - (A) 1.0 mole of H<sub>2</sub>O is produced
     (B) 2.0 moles of NO will be produced
  - (C) All the oxygen will be consumed
  - (D) All the ammonia will be consumed.

# **National Science Olympiad Sample Papers**

**GUPTA SHIKHA** 

### **National Science Olympiad Sample Papers:**

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Handbook is prepared for providing some additional study materials to fellow students of Class X of the National Curriculum and State Boards Most of the questions were adopted from the previous year question papers of different boards and duly presented in the form of different worksheets Topics covered 1 Biological processes 2 Reproduction in Plants and Animals 3 Genetics and Evolution 4 Physiology of Hearing and Vision For additional practice questions check out the Extended Study Modules by exploring the public domains Chandan Sukumar Sengupta You can use them to study on internet your smartphone tablet or computer anytime anywhere **National Science Olympiad - Class 9 (With OMR Sheets)** AGARWAL PREETI, 2016-04-20 The book National Science Olympiad has been divided into five sections namely Science Logical Reasoning Achievers section Subjective section Model Papers In every chapter the theory has been explained through solved examples illustrations and diagrams wherever required Multiple Choice Questions MCQs with detailed solutions are provided in the end of each chapter to enhance the problem solving skills of candidates The questions in the Achievers section are set to evaluate scientific skills of brilliant students whereas the Subjective section includes questions of descriptive nature Two Model Papers have been included to do practice on the expected questions A CD containing Study Chart for systematic preparation Tips Tricks to crack Science Olympiad Pattern of exam and links of Previous Years Papers is accompanied with this book The book is also useful for various competitive exams such as NTSE NSTSE and SLSTSE as well Science Course Improvement Projects National Science Foundation (U.S.),1962 v spublishers ,1999 Olympiad Sample Paper 9 Editorial Board, 2018-05-18 The Olympiad sample papers have been developed by experts in their respective fields to make students familiar with the syllabus covered in the exam and the question asking pattern followed by the marking scheme Set on the lines of MCO Multiple Choice Questions format adopted in the exam there are two sets of papers on each of Mathematics Science Cyber and English Olympiads for Class 9 Answers keys are given to enable students to verify the correctness of the answers Where necessary steps to solving questions are also given Students can practice through these papers check their scores and assess their level of preparedness and knowledge This kind of meticulous attention to detail is sure to help them make a smart plan and strategy for preparation of these challenging NCO NSO IEO and IMO exams From the sample papers students will get a fair idea about the type of questions asked in the examination In this series we present for students a full range of sample papers from Class 1st to 10th Syllabus question patterns and marking arrangements are given so that the student can learn and prepare for the exam accordingly These sample papers will prove to be of premier importance while preparing for the Olympiad exams Olympiad Sample Paper 1 Editorial Board, 2018-05-10 Strategies for Winning Science Fair Projects Joyce Henderson, Heather Tomasello, 2001-11-06 Discover the Secrets of Science Fair Success with This Essential Guide Written by a science fair judge and an international science fair winner this must have resource is packed with strategies and pointers for putting together a winning science fair project Here you ll get the nitty gritty on a wide variety of topics from the fundamentals of the science

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# **Table of Contents National Science Olympiad Sample Papers**

- 1. Understanding the eBook National Science Olympiad Sample Papers
  - The Rise of Digital Reading National Science Olympiad Sample Papers
  - Advantages of eBooks Over Traditional Books
- 2. Identifying National Science Olympiad Sample Papers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an National Science Olympiad Sample Papers
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from National Science Olympiad Sample Papers
  - Personalized Recommendations
  - National Science Olympiad Sample Papers User Reviews and Ratings
  - National Science Olympiad Sample Papers and Bestseller Lists

- 5. Accessing National Science Olympiad Sample Papers Free and Paid eBooks
  - National Science Olympiad Sample Papers Public Domain eBooks
  - National Science Olympiad Sample Papers eBook Subscription Services
  - National Science Olympiad Sample Papers Budget-Friendly Options
- 6. Navigating National Science Olympiad Sample Papers eBook Formats
  - o ePub, PDF, MOBI, and More
  - National Science Olympiad Sample Papers Compatibility with Devices
  - National Science Olympiad Sample Papers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of National Science Olympiad Sample Papers
  - Highlighting and Note-Taking National Science Olympiad Sample Papers
  - Interactive Elements National Science Olympiad Sample Papers
- 8. Staying Engaged with National Science Olympiad Sample Papers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers National Science Olympiad Sample Papers
- 9. Balancing eBooks and Physical Books National Science Olympiad Sample Papers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection National Science Olympiad Sample Papers
- 10. Overcoming Reading Challenges
  - o Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine National Science Olympiad Sample Papers
  - Setting Reading Goals National Science Olympiad Sample Papers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of National Science Olympiad Sample Papers
  - Fact-Checking eBook Content of National Science Olympiad Sample Papers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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