

### 3. Marking points

#### 3.1 Marking of lists

This applies to questions requiring a set number of responses, but for which students have provided extra responses. The general principle to be followed in such a situation is that 'right + wrong = wrong'.

Each error / contradiction negates each correct response. So, if the number of error / contradictions equals or exceeds the number of marks available for the question, no marks can be awarded.

However, responses considered to be neutral (indicated as \* in example 1) are not penalised.

Example 1: What is the pH of an acidic solution?

[1 mark]

Student	Response	Marks awarded
1	green, 5	0
2	red*, 5	1
3	red*, 8	0

Example 2: Name two planets in the solar system.

[2 marks]

Student	Response	Marks awarded
1	Mercury, Venus, Earth, Mars, Jupiter, Saturn, Uranus, Neptune	0

#### 3.2 Use of chemical symbols/formulae

For questions asking for chemical symbols/formulae instead of a named chemical name, full credit can be given if the student formula is correct and is in the correct format of the question. Such errors as 'hydroxide' are not penalised.

#### 3.3 Marking procedures for calculations

Errors should be reported for each stage of the calculation completed correctly. An incorrectly calculated value does not prevent full credit for the correct use of formulae to give the correct numerical answer. Different rounding errors.

#### 3.4 Interpretation of %

Answers using the word 'of' should be given credit only if it is clear that they 'of' refers to the correct subject.

# Jan 13 Chem 5 Mark Scheme

**Camilla Rothe**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name Camilla Rothe.

### Jan 13 Chem 5 Mark Scheme:

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