

12.3 Limiting Reagent and Percent Yield



BY:
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Reaction Stoichiometry

Amount of Product Formed and Percent Yield



Theoretical Yield: 0.074 g O₂

Actual Yield: 0.069 g O₂

Percent Yield: $\frac{0.069 \text{ g O}_2}{0.074 \text{ g O}_2} = 93.2\% \text{ Yield}$

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Murphy, Patrick Woodward, Steven
Langford, Dalius Sagatys, Adrian
George**

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