LIMITING REACTANTS

1. What mass of CS₂ is produced when 17.5 g of C are reacted with 39.5 g of SO₂ according to the equation below?

$$5C+2SO_2 \longrightarrow CS_2+4CO$$

What mass of the excess reactant will be left over?

2. What mass of NO is produced when 87.0 g of Cu are reacted with 225 g of HNO₃ according to the equation?

$$3Cu + 8HNO_3 \longrightarrow 3Cu(NO_3)_2 + 2NO + 4H_2O$$

What mass of the excess reactant will be left over?

3. What mass of P_4 is produced when 41.5 g of $Ca_3(PO_4)_2$, 26.5 g of SiO_2 and 7.80 g of C are reacted according to the equation

$$2 C a_3 |PO_4|_2 + 6 SiO_2 + 10 C \longrightarrow P_4 + 6 CaSiO_3 + 10 CO$$

How many grams of each excess reactant will remain unreacted?

<u>Limiting Reaactants And Percent Yeild 9 3 Section</u> <u>Review</u>

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