#### Chapter 2 The Nature of Matter - Answers

#### 2.1 Properties of Matter

#### Warm Up, p. 60

- 1. For example: cold, white, crystalline, slippery, compressible, cohesive
- 2. For example: light, heavy, fluffy, wet
- For example: the air temperature and pressure

#### Quick Check, p. 62

- 1. The stuff that materials are composed of
- 2. A quality of a thing, especially a quality common to a group, type, class, etc.
- A quality that is or depends upon the amount of the material
- A property that describes a chemical change. That is one in which a new substance(s) or species is formed, or otherwise describes the tendency of a chemical to react

#### Quick Check, p. 63

- The average mechanical energy of the particles that compose a material
- 2. The total mechanical energy of an object's or a material's particles
- 3. The energy transferred from one body to another because of a difference in temperature

#### Quick Check, p. 65

- 1. The process of changing from a solid to a liquid
- The vigorous bubbling that occurs within the body of a liquid as it vaporizes internally
- The amount of heat required to melt a specified amount of a substance at its melting point

#### Quick Check, p. 67

- A property of a solution that only depends on the concentration of the dissolved particles, not on their identity
- For example: boiling point elevation
- 3. A solution of a metal and another solid or solids

# **Nature Of Matter Answers Key**

**Verne Hobson Booth** 

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