

$$k = \frac{5}{3}$$

T_2 can be calculated by applying the polytropic relation derived for adiabatic expansions.
From Equation 2.89

$$\begin{aligned} pV^k &= \text{const} \\ \therefore pV^k &= \text{const} \end{aligned}$$

The application of the above equation

$$p_1 V_1^k = p_2 V_2^k$$

Since T is a constant, substitution of the expression for T into the polytropic relation results in

$$\frac{p_1 T_1}{p_1 V_1^k} = \frac{p_2 T_2}{p_2 V_2^k}$$

This relation can be used to calculate the T_2

$$T_2 = T_1 \left(\frac{V_1}{V_2} \right)^{k-1}$$

Since that T_1 is known, values of V_1 and V_2 can be obtained

$$V = \frac{RT}{p} = \frac{RT}{p_1} \left(\frac{p_1}{p_2} \right)^{1/k}$$

Koretsky Thermodynamics 2nd Edition Solution Manual

**Joe M. Smith, Hendrick C. Van
Ness, Michael M. Abbott, John J. Hwalek**



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