

# CBSE PAPER-2024 (SET-1)

## CHEMISTRY

1. The molar ionic conductivities of  $\text{Ca}^{2+}$  and  $\text{Cl}^-$  are  $119.0$  and  $76.3 \text{ S cm}^2 \text{ mol}^{-1}$  respectively. The value of limiting molar conductivity of  $\text{CaCl}_2$  will be : [1]  
 (a)  $195.3 \text{ S cm}^2 \text{ mol}^{-1}$  (b)  $43.3 \text{ S cm}^2 \text{ mol}^{-1}$   
 (c)  $314.3 \text{ S cm}^2 \text{ mol}^{-1}$  (d)  $271.6 \text{ S cm}^2 \text{ mol}^{-1}$
2. Consider the following reaction : [1]



Identify A and B from the given options :

- (a) A – Methanol, B – Potassium formate (b) A – Ethanol, B – Potassium formate  
 (c) A – Methanal, B – Ethanol (d) A – Methanol, B – Potassium acetate
3. Which of the following acids represents Vitamin C ? [1]  
 (a) Saccharic acid (b) Gluconic acid  
 (c) Ascorbic acid (d) Benzoic acid
4. Rosenmund reduction is used for the preparation of Aldehydes. The catalyst used in this reaction is : [1]  
 (a)  $\text{Pd} - \text{BaSO}_4$  (b) Anhydrous  $\text{AlCl}_3$   
 (c) Iron (III) oxide (d)  $\text{HgSO}_4$
5. Which alkyl halide from the given options will undergo  $\text{S}_{\text{N}}1$  reaction faster ? [1]  
 (a)  $(\text{CH}_3)_3\text{C}-\text{Br}$  (b)  $(\text{CH}_3)_2\text{CH}-\text{Br}$  (c)  $\text{CH}_3-\text{CH}_2-\text{Br}$  (d)  $(\text{CH}_3)_3\text{C}-\text{CH}_2-\text{Br}$
6. From the elements of 3d series given below, which element shows the maximum number of oxidation states ? [1]  
 (a) Scandium (b) Manganese (c) Chromium (d) Titanium

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**Jim Clark**



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## **Table of Contents Lgcse Edexcel Chemistry Paper1 January 2014**

1. Understanding the eBook Lgcse Edexcel Chemistry Paper1 January 2014
  - The Rise of Digital Reading Lgcse Edexcel Chemistry Paper1 January 2014
  - Advantages of eBooks Over Traditional Books
2. Identifying Lgcse Edexcel Chemistry Paper1 January 2014
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Lgcse Edexcel Chemistry Paper1 January 2014
  - User-Friendly Interface
4. Exploring eBook Recommendations from Lgcse Edexcel Chemistry Paper1 January 2014
  - Personalized Recommendations
  - Lgcse Edexcel Chemistry Paper1 January 2014 User Reviews and Ratings
  - Lgcse Edexcel Chemistry Paper1 January 2014 and Bestseller Lists
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  - Lgcse Edexcel Chemistry Paper1 January 2014 eBook Subscription Services
  - Lgcse Edexcel Chemistry Paper1 January 2014 Budget-Friendly Options
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- ePub, PDF, MOBI, and More
  - Lgcse Edexcel Chemistry Paper1 January 2014 Compatibility with Devices
  - Lgcse Edexcel Chemistry Paper1 January 2014 Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Lgcse Edexcel Chemistry Paper1 January 2014
  - Highlighting and Note-Taking Lgcse Edexcel Chemistry Paper1 January 2014
  - Interactive Elements Lgcse Edexcel Chemistry Paper1 January 2014
8. Staying Engaged with Lgcse Edexcel Chemistry Paper1 January 2014
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Lgcse Edexcel Chemistry Paper1 January 2014
9. Balancing eBooks and Physical Books Lgcse Edexcel Chemistry Paper1 January 2014
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Lgcse Edexcel Chemistry Paper1 January 2014
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Lgcse Edexcel Chemistry Paper1 January 2014
- Setting Reading Goals Lgcse Edexcel Chemistry Paper1 January 2014
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Lgcse Edexcel Chemistry Paper1 January 2014
- Fact-Checking eBook Content of Lgcse Edexcel Chemistry Paper1 January 2014
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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