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Chemistry 11

Classifying Matter

Assignment

1. Complete the following sentences by filling in the appropriate word from the list below.

Elements
Mixtures
Compound
Symbols
Properties

Carbon
Pure
Letters
Carbon dioxide

- a. _____ Elements _____ cannot be separated by physical or chemical means.
- b. Elements are abbreviated with _____ symbols _____, which consist of one or two _____ letters _____.
- c. Two or more elements chemically combined make up a _____ compound _____.
- d. _____ Carbon _____ is an example of an element and _____ carbon dioxide _____ is an example of a compound.
- e. Elements and compounds are called _____ pure _____ substances because they have a unique set of chemical and physical _____ properties _____.

2. Fill in the blanks in the following chart.

Element Name	Element Symbol	Common Ions Formed
a. Sodium	<u>Na</u>	<u>+1</u>
b. Copper	Cu	<u>+2, +1</u>
c. Tungsten	W	<u>+6</u>
d. Iron	<u>Fe</u>	<u>+3, +2</u>
e. Tin	Sn	<u>+4, +2</u>
f. Potassium	<u>K</u>	<u>+1</u>
g. Gold	Au	<u>+3, +1</u>
h. Silver	Ag	<u>+1</u>

3. Classify each of the following as one of an Atom (A), a Molecule (M) or an Ion (I).

a. P^{3-} I

b. H_2O M

c. O_2 M

d. Au^{3+} I

e. Ge A

f. O A

g. Ca^{2+} I

h. NH_3 M

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Richard Bailey



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