

Limiting and Excess Reactants

Answers

- ① For the following equation, determine which reactant is limiting and which reactant is in excess. The amount of reagents used is shown.



$$40 \text{ g Fe} \times \frac{1 \text{ mol Fe}}{55.8 \text{ g Fe}} \times \frac{1 \text{ mol Fe}_3\text{O}_4}{3 \text{ mol Fe}} = 0.239 \text{ mol Fe}_3\text{O}_4$$

$$16 \text{ g H}_2\text{O} \times \frac{1 \text{ mol H}_2\text{O}}{18 \text{ g H}_2\text{O}} \times \frac{1 \text{ mol Fe}_3\text{O}_4}{4 \text{ mol H}_2\text{O}} = 0.222 \text{ mol Fe}_3\text{O}_4$$

0.222 mol is less than 0.239 mol. Therefore, the limiting reactant is H₂O, and the excess reactant is Fe.

- ② 35.5 g of silver nitrate is reacted with 35.5 grams of sodium sulfide, producing silver sulfide and sodium nitrite.

- (a) Write and balance the equation.



- (b) Calculate the number of grams of silver sulfide produced in grams.

$$35.5 \text{ g AgNO}_3 \times \frac{1 \text{ mol AgNO}_3}{153.9 \text{ g AgNO}_3} \times \frac{1 \text{ mol Ag}_2\text{S}}{2 \text{ mol AgNO}_3} = 0.115 \text{ mol Ag}_2\text{S}$$

$$35.5 \text{ g Na}_2\text{S} \times \frac{1 \text{ mol Na}_2\text{S}}{78 \text{ g Na}_2\text{S}} \times \frac{1 \text{ mol Ag}_2\text{S}}{2 \text{ mol Na}_2\text{S}} = 0.455 \text{ mol Ag}_2\text{S}$$

The amount of silver sulfide produced is 0.115 mol. Let us convert this into grams.

$$0.115 \text{ mol Ag}_2\text{S} \times \frac{247.8 \text{ g Ag}_2\text{S}}{1 \text{ mol Ag}_2\text{S}} = 28.5 \text{ g Ag}_2\text{S}$$

- ③ For the following reaction: $\text{C}_3\text{H}_8 + 5 \text{ O}_2 \rightarrow 3 \text{ CO}_2 + 4 \text{ H}_2\text{O}$

- (a) Write and balance the equation.

$$0.37 \text{ g CO}_2 \times \frac{1 \text{ mol of CO}_2}{44.01 \text{ g CO}_2} \times \frac{1 \text{ mol C}_3\text{H}_8}{3 \text{ mol CO}_2} = 0.00280 \text{ mol C}_3\text{H}_8$$

- (b) How many grams of propane is this.

$$0.00280 \text{ mol C}_3\text{H}_8 \times \frac{44.1 \text{ g C}_3\text{H}_8}{1 \text{ mol C}_3\text{H}_8} = 0.12348 \text{ g C}_3\text{H}_8$$

Limiting Reactants Study Guide Answers

DJ Losen



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Limiting Reactants Study Guide Answers: Bestsellers in 2023 The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous captivating novels captivating the hearts of readers worldwide. Lets delve into the realm of bestselling books, exploring the fascinating narratives that have charmed audiences this year. The Must-Read : Colleen Hoover's "It Ends with Us" This poignant tale of love, loss, and resilience has gripped readers with its raw and emotional exploration of domestic abuse. Hoover skillfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can succeed. Uncover the Best : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This spellbinding historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids absorbing storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Discover the Magic : Delia Owens "Where the Crawdads Sing" This captivating coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens weaves a tale of resilience, survival, and the transformative power of nature, entrancing readers with its evocative prose and mesmerizing setting. These top-selling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of compelling stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a exceptional and gripping novel that will keep you wondering until the very end. The novel is a warning tale about the dangers of obsession and the power of evil.

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