

# Matric Final Exam Time Table 2025

## South Africa Page 2



<b>WEEK 4</b>	<b>09:00</b>	<b>14:00</b>
Monday 10/11	Physical Sciences (Chemistry) P2 (3hrs) Technical Sciences P2 (1½hrs)	Visual Arts P1 (3hrs) Nautical Science P1 (3hrs)
Tuesday 11/11	Afrikaans HL P1 (2hrs) Afrikaans FAL P1 (2hrs) Afrikaans SAL P1 (2hrs)	Business Studies P2 (2hrs)
Wednesday 12/11	Geography (Climate and Weather, Geomorphology and Map Work) P1 (3hrs)	Computer Applications Tech P2 (Theory) (3hrs) Nautical Science P2 (3hrs)
Thursday 13/11	English HL P2 (2½hrs) English FAL P2 (2½hrs) English SAL P2 (1½hrs)	Information Technology P2 (Theory) (3hrs) Marine Sciences P1 (2½hrs)
Friday 14/11	Life Sciences P1 (2½hrs)	Electrical Technology (3hrs)
<b>WEEK 5</b>	<b>09:00</b>	<b>14:00</b>
Monday 17/11	Life Sciences P2 (2½hrs)	Dramatic Arts (3hrs)
Tuesday 18/11	Geography (Rural and Urban Settlements, Economic Geography of SA and Map Work) P2 (3hrs)	Engineering Graphics and Design P2 (3hrs)
Wednesday 19/11	Economics P1 (2hrs)	Tourism (3hrs)
Thursday 20/11	Sepedi, Sesotho, Setswana, Xitsonga, Tshivenda HL P2 (2½hrs), FAL P2 (2½hrs), SAL P2 (1½hrs) South African Sign Language HL P2 (2½hrs)	Agricultural Sciences P2 (2½hrs)
Friday 21/11	Afrikaans HL P2 (2½hrs) Afrikaans FAL P2 (2½hrs) Afrikaans SAL P2 (1½hrs)	Economics P2 (2hrs)
<b>WEEK 6</b>	<b>09:00</b>	<b>14:00</b>
Monday 24/11	Agricultural Technology (3hrs) Music P1 Theory (3hrs)	Design (3hrs)
Tuesday 25/11	Consumer Studies (3hrs) Hospitality Studies (3hrs)	Dance P1 (3hrs)
Wednesday 26/11	Agricultural Management Practices (3hrs) Marine Sciences P2 (2½hrs)	Music P2 Comprehension (1½hrs)
Thursday 27/11	<b>CAT P1 rewrite</b> (3hrs) Practical <b>IT P1 rewrite</b> (3hrs) Practical	

	<b>09:00</b>
Friday, 17 <sup>th</sup> October 2025	Pledge Signing
Monday, 1 <sup>st</sup> September 2025	Life Orientation (LO CAT) (2½hrs)
Tuesday, 14 <sup>th</sup> October 2025	Life Orientation (LO CAT) (2½hrs) Rewrite
14 <sup>th</sup> August 2025 to 24 <sup>th</sup> October 2025	Performing Arts Practical
14 <sup>th</sup> October 2025 to 24 <sup>th</sup> October 2025	Visual Arts and Design Practical

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Enquiries: Dr RR Poliah

Chief Directorate: National Assessment and Public Examinations (NAPE)  
012 357 3900



# Limpopo Final Time Table 2014

**L Darling-Hammond**



### **Limpopo Final Time Table 2014:**

*The Geography of South Africa* Jasper Knight, Christian M. Rogerson, 2018-10-03 This edited collection examines contemporary directions in geographical research on South Africa. It encompasses a cross section of selected themes of critical importance not only to the discipline of Geography in South Africa but also of relevance to other areas of the Global South. All chapters are original contributions providing a state of the art research baseline on key themes in physical human and environmental geography and in understanding the changing geographical landscapes of modern South Africa. These contributions set the scene for an understanding of the relationships between modern South Africa and the wider contemporary world including issues of sustainable development and growth in the Global South. *Tuberculosis, Leprosy and Mycobacterial Diseases of Man and Animals* Harshini Mukundan, Mark Chambers, Ray Waters, Michelle

Larsen, 2015-09-28 Mycobacteria are bacterial pathogens which cause diseases in humans and non human animals. This monograph will primarily cover the most important and widely researched groups of mycobacteria members of the Mycobacterium tuberculosis complex (MTC) and Mycobacterium leprae across a wide range of host species. M. tuberculosis and M. bovis are particularly relevant with the increasing drug resistance and co-infection with HIV associated with M. tuberculosis and the possible cross-infection of badgers and cattle associated with M. bovis. This book will provide a reference for researchers working in different fields, creating a work which draws together information on different pathogens and by considering the diseases in a zoonotic context provides a One Health approach to these important groups of diseases. **The**

**Railway Gazette**, 1962      **Land News**, 2010      **Railway Gazette. [Weekly]**, 1962      **Outlook on Agriculture**, 2015

*Papers R.P.*, 1967

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## **Table of Contents Limpopo Final Time Table 2014**

1. Understanding the eBook Limpopo Final Time Table 2014
  - The Rise of Digital Reading Limpopo Final Time Table 2014
  - Advantages of eBooks Over Traditional Books
2. Identifying Limpopo Final Time Table 2014
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Limpopo Final Time Table 2014
  - User-Friendly Interface
4. Exploring eBook Recommendations from Limpopo Final Time Table 2014
  - Personalized Recommendations
  - Limpopo Final Time Table 2014 User Reviews and Ratings
  - Limpopo Final Time Table 2014 and Bestseller Lists
5. Accessing Limpopo Final Time Table 2014 Free and Paid eBooks

- Limpopo Final Time Table 2014 Public Domain eBooks
- Limpopo Final Time Table 2014 eBook Subscription Services
- Limpopo Final Time Table 2014 Budget-Friendly Options
- 6. Navigating Limpopo Final Time Table 2014 eBook Formats
  - ePub, PDF, MOBI, and More
  - Limpopo Final Time Table 2014 Compatibility with Devices
  - Limpopo Final Time Table 2014 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Limpopo Final Time Table 2014
  - Highlighting and Note-Taking Limpopo Final Time Table 2014
  - Interactive Elements Limpopo Final Time Table 2014
- 8. Staying Engaged with Limpopo Final Time Table 2014
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Limpopo Final Time Table 2014
- 9. Balancing eBooks and Physical Books Limpopo Final Time Table 2014
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Limpopo Final Time Table 2014
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Limpopo Final Time Table 2014
  - Setting Reading Goals Limpopo Final Time Table 2014
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Limpopo Final Time Table 2014
  - Fact-Checking eBook Content of Limpopo Final Time Table 2014
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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subsequent  $\Delta h$  values  $2 \text{ CO}_2 \text{ g}$   $\text{H}_2\text{O g}$   $\text{C}_2\text{H}_2 \text{ g}$   $5 \text{ O}_2 \text{ g}$   $2 \text{ H}_2 \text{ g}$   $\text{C}_2\text{H}_6 \text{ g}$   $\Delta h$   $94.5 \text{ kJ}$   $\text{H}_2\text{O g}$   $\text{H}_2 \text{ g}$   $\frac{1}{2} \text{ O}_2 \text{ g}$   $\Delta h$   $71.2 \text{ kJ}$   $\text{C}_2\text{H}_6 \text{ g}$   $7 \text{ O}_2 \text{ g}$   $2 \text{ CO}_2 \text{ g}$   $3 \text{ H}_2 \text{O g}$   $\Delta h$   $283 \text{ kJ}$   $\text{C}_2\text{H}_6 \text{ g}$

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web sep 22 2021 hess s law states that the enthalpy change of an overall process is equal to the sum of the enthalpy changes of its individual steps hess s law example 12.1 determine  $\Delta h$  for the target reaction  $\text{C}_2\text{NO}_2 \text{ g}$   $1 \text{ O}_2 \text{ g}$   $\text{N}_2\text{O}_5 \text{ g}$  given the following information

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web libretexts states hess s law states that regardless of the multiple stages or steps of a reaction the total enthalpy change for the reaction is the sum of all changes libretexts quot hess s law quot 2019 which determines the enthalpy change value hess s law is written as  $\Delta h_{\text{target}} = \Delta h_1 + \Delta h_2 + \Delta h_3$  or  $\Delta h_{\text{target}} = \sum \Delta h_{\text{known}}$

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web hess s law states that if a process can be expressed as the sum of two or more steps the enthalpy change for the overall process is the sum of the  $\Delta h$  values for each step to use hess s law two principles must be understood one if an equation is reversed the sign of the  $\Delta h$  value is also reversed two if an equation is multiplied by a

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web hess s law ws 6a answer key directions for each of the following problems find the unknown  $\Delta h$  and show the reactions adding up to the overall reaction

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web hess s law labs by austin lee alayna baron lily zmachinski introduction in order to calculate the enthalpy change for the combustion of magnesium oxide  $\text{Mg s}$   $1 \text{ O}_2 \text{ g}$   $\text{MgO s}$  we used

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web hess s law states that the total enthalpy change for the reaction will be the sum of all those changes no matter how many different steps or stages in the reaction there are cohen 2016 the equations for the reactions in the experiment done are as follows  $1 \text{ NaOH s}$   $\text{Na aq}$   $\text{OH aq}$

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web hess s laws abstract the following experiment was designed using hess s law and simple calorimetry to determine the

enthalpy for naoh s hess s law says that the overall change in enthalpy for the reaction is the sum of all changes regardless of the different stages or phases of a reaction

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web hess s law states that if a reaction is the sum of two or more other reactions the  $\Delta h$  for the overall process must be the sum of the  $\Delta h$  values of the constituent reactions enthalpy change  $\Delta h$  is independent of the path that a reaction follows to move from reactants

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web this activity provides a demonstration of hess law using three reactions the solubility naoh in water the solubility naoh in hcl and the reaction of a solution of hcl and a solution of naoh

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web hess s law worksheet answers 1 calculate  $\Delta H$  for the reaction  $C_2H_4(g) + H_2(g) \rightarrow C_2H_6(g)$  from the following data  $C_2H_4(g) + 3 O_2(g) \rightarrow 2 CO_2(g) + 2 H_2O(l) \quad \Delta H = -1411 \text{ kJ}$   $C_2H_6(g) + \frac{7}{2} O_2(g) \rightarrow 2 CO_2(g) + 3 H_2O(l) \quad \Delta H = -1560 \text{ kJ}$

*7a first law enthalpy calorimetry and hess s law worksheet* - Oct 05 2022

web mar 13 2023 be able to use hess s law to calculate reaction enthalpies first law of thermodynamics energy is the ability to do work or transfer heat work is the transfer of energy from one body to another in a sense work is energy in the process of transfer

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web may 8 2015 hess law lab by maya parks partners ben seufert kelsea floyd convert joules to kj in your final answer multiply the mass by the change in temperature and the c given trial 1 use hess s law and the following equations and  $\Delta H$  values to determine the heat of reaction for the reaction  $C_2H_4$

**experiment 9 enthalpy of reaction hess s law** - Jul 14 2023

web thus for reaction 9 2 in which 1 mole of  $H_2SO_4$  and 1 mole of  $OH^-$  combine to form 2 moles of  $H_2O$   $\Delta H = -55.8 \text{ kJ}$  as shown note that the enthalpy change  $\Delta H_{\text{products}} - \Delta H_{\text{reactants}}$  is positive if heat is absorbed that is if  $\Delta H_{\text{products}} > \Delta H_{\text{reactants}}$

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web hess s law enthalpy of reaction and hess s law abstract the goal of this lab is to perform experiments and practice hess s law by determining enthalpy change in chemical reactions in this lab three experiments were conducted in order to test hess s law the experiments required observing and collecting data on three different reactions

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web objective the purpose of the lab is to combine two reactants in the calorimeter and measure the heats of reaction in solution eventually allowing an individual to prove or disprove hess s law introduction thermodynamics is the study of energy

exchanged between the system and the surroundings

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web 3 12 hess s law in this experiment you will measure the amount of heat released in these three related exothermic reactions

1  $\text{NaOH (s)} \rightarrow \text{Na}^+ \text{(aq)} + \text{OH}^- \text{(aq)}$   $\Delta H_1$

2  $\text{NaOH (s)} + \text{H}^+ \text{(aq)} \rightarrow \text{Na}^+ \text{(aq)} + \text{H}_2\text{O (l)}$   $\Delta H_2$

3  $\text{NaOH (s)} + \text{HCl (aq)} \rightarrow \text{Na}^+ \text{(aq)} + \text{Cl}^- \text{(aq)} + \text{H}_2\text{O (l)}$   $\Delta H_3$