

Limiting Reactant and Percent Yield Worksheet

(Show your work)

1. Consider the following reaction: $2 \text{Al} + 6 \text{HBr} \rightarrow 2 \text{AlBr}_3 + 3 \text{H}_2$

When 3.22 moles of Al reacts with 6.96 moles of HBr, what are the limiting and excess reactants?

2. Consider the following reaction: $4 \text{FeS}_2 + 11 \text{O}_2 \rightarrow 2 \text{Fe}_2\text{O}_3 + 8 \text{SO}_2$

When 26.62 moles of FeS_2 reacts with 59.44 moles of O_2 , what are the limiting and excess reactants?

3. Consider the following reaction: $3 \text{Si} + 2 \text{N}_2 \rightarrow \text{Si}_3\text{N}_4$

When 600 g of Si reacts with 500 g of N_2 , What are the limiting and excess reactants?

4. Given the following equation: $\text{Al}_2(\text{SO}_4)_3 + 6 \text{NaOH} \rightarrow 3 \text{Na}_2\text{SO}_4 + 2 \text{Al}(\text{OH})_3$

If 10.0 g of $\text{Al}_2(\text{SO}_4)_3$ is reacted with 10.0 g of NaOH, determine the limiting and excess reactants.

Limiting Reactant And Percent Yield Review Sheet 3

Answers

Holder



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Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03
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Guide to accompany Chemistry Martin Silberberg,Libby Weberg,2005-01-06 **General Chemistry Study Guide Sixth Edition** Darrell D. Ebbing,1999 **Study Guide to Accompany Basics for Chemistry** Martha Mackin,1982-01-01 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical

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