

Limiting Reagent and Percentage Yield

- 1 Dichlorine monoxide is produced by passing chlorine gas over heated mercury (II) oxide according to the following unbalanced reaction:



Balance the reaction. Suppose the quantity of the reactants is sufficient to produce 0.86 g of Cl_2O , but only 0.71 g is obtained. What is the percentage yield?

- 2 Consider the following reaction:



The theoretical yield of beryllium chloride was 10.7 g. Suppose the reaction actually yields 4.5 g. What is the percentage yield?

- 3 Nitrogen and hydrogen at high temperatures are converted to ammonia using the following reaction:



When 400 g of H_2 are added to an excess amount of N_2 , 104 g of NH_3 are formed. Calculate the percentage yield.

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Chemistry for the Utterly Confused John T. Moore,Richard H. Langley,2007-06-01 Banish bafflement in this tough subject From formulas and lab techniques to the periodic table Chemistry for the Utterly Confused focuses on the areas of maximum confusion and breaks down the most difficult chemistry topics into easy to understand concepts This invaluable guide also teaches problem solving skills you need to master this imposing subject Whether you re in high school in college or simply brushing up on chemistry knowledge this fun easily accessible book will make understanding chemistry a breeze

Chemistry John Olmsted,Gregory M. Williams,1997 Textbook outlining concepts of molecular science

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Ebook: Chemistry: The Molecular Nature of Matter and Change Silberberg,2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change

Chemistry John A. Olmsted,Gregory M. Williams,Robert Charles Burk,2016-01-14 Olmsted Burk is an introductory

general chemistry text designed specifically with Canadian professors and students in mind A reorganized Table of Contents and inclusion of SI units IUPAC standards and Canadian content designed to engage and motivate readers distinguish this text from many of the current text offerings It more accurately reflects the curriculum of most Canadian institutions Instructors will find the text sufficiently rigorous while it engages and retains student interest through its accessible language and clear problem solving program without an excess of material that makes most text appear daunting and redundant

General Chemistry, Reactions First Kevin Revell, 2024-12-04 Revell's General Chemistry empowers students to grasp essential topics and concepts with more ease Using a friendly approach the text uses metaphors and relatable examples to demystify even the most challenging subjects in general chemistry

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Introductory Chemistry Math Review Tro, Sara Long, 2005-02 This print resource packaged with the text reinforces the skills necessary to succeed in chemistry Keyed specifically to chapters in Introductory Chemistry Second Edition it includes additional mathematics review problem solving tools and examples

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Student Study Guide to Accompany Petrucci's General Chemistry Robert K. Wismer, 1985

Understanding Chemistry Fred M. Dewey, 1994

Student Study Guide to Accompany Petrucci's General Chemistry, 3rd. Ed Robert K. Wismer, 1982

Prentice Hall Chemistry, 2000

5 Steps to a 5: AP Chemistry 2021 Mary Millhollon, Richard H. Langley, 2020-10-02

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