

Spring Semester Final Exam Study Guide

Chemistry Advanced

Name _____

Period _____

Naming and Formula Writing

1. Write the name or formula for each of the following:

AlCl_3	HClO_2	Mercury (II) bromide
Cu_2S	$(\text{NH}_4)_2\text{SO}_4$	Phosphorous acid
NaCN	I_4O_{10}	Nitric acid
HI	H_3N	Sodium nitrite
PCl_3	NiN	Magnesium hydroxide
CrBr_3	H_3PO_4	Carbon trisulfide

Chemical Equations

2. State the law of conservation of matter.



3. Using the chemical equation above:

- Underline the subscripts
- Box the coefficients (**bold**)
- Identify the diatomic molecules (**arrows**)

4. Balance the following equations and identify the type of reaction (synthesis, decomposition, single replacement, double replacement, combustion)



Language Arts Spring Final Study Guide

M Tight



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