

**1** Generally transition elements and their salts are coloured due to the presence of unpaired electrons in metal ions. Which of the following compounds are coloured?

- KMnO<sub>4</sub>
- Ce(SO<sub>4</sub>)<sub>2</sub>
- TiCl<sub>3</sub>
- Cu<sub>2</sub>Cl<sub>2</sub>

Choose the correct option

- (a, b)
- (b, c)
- (c, d)
- (a, d)

**2** General electronic configuration of actinoids is (n-2)f<sup>1-14</sup>(n-1)d<sup>0-2</sup>ns<sup>2</sup>. Which of the following actinoids have one electron in 6d orbital?

- U (Atomic no. 92)
- Np (Atomic no. 93)
- Pu (Atomic no. 94)
- Am (Atomic no. 95)

Choose the correct option

- (a, b)
- (b, c)
- (c, d)
- (a, d)

**3** Which of the following lanthanoids show +2 oxidation state besides the characteristic oxidation state +3 of lanthanoids?

- Ce
- Eu
- Yb
- Ho

Choose the correct option

- (a, b)
- (b, c)
- (c, d)
- (a, d)

**4** On heating lead nitrate forms oxides of nitrogen and lead. The oxides formed are-

- N<sub>2</sub>O, PbO
- NO<sub>2</sub>, PbO
- NO, PbO
- NO, PbO<sub>2</sub>

**5** A brown ring is formed in the ring test for NO<sub>3</sub><sup>-</sup> ion. It is due to the formation of-

- [Fe (H<sub>2</sub>O)<sub>5</sub> (NO)]<sup>2+</sup>
- FeSO<sub>4</sub>, NO<sub>2</sub>
- [Fe (H<sub>2</sub>O)<sub>4</sub> (NO)<sub>2</sub>]<sup>2+</sup>
- FeSO<sub>4</sub>, HNO<sub>3</sub>

**6** The oxidation state of the central atom in the anion of the compound NaH<sub>2</sub>PO<sub>2</sub> will be-

- +3
- +5
- +1
- 3

**7** A black compound of manganese reacts with a halogen acid to give greenish-yellow gas. When an excess of this gas reacts with NH<sub>3</sub> an unstable trihalide is formed. In this process, the oxidation state of nitrogen changes from-

- 3 to +3
- 3 to 0
- 3 to +5
- 0 to -3

**8** Which of the following is isoelectronic pair?

- ICl<sub>2</sub>, ClO<sub>2</sub>
- BrO<sub>2</sub><sup>-</sup>, BrF<sub>2</sub><sup>+</sup>
- ClO<sub>2</sub>, BrF
- CN<sup>-</sup>, O<sub>3</sub>

**9** Which of the following statements are correct for SO<sub>2</sub> gas?

- It acts as a bleaching agent in moist conditions.
  - Its molecule has linear geometry.
  - Its dilute solution is used as a disinfectant.
  - It can be prepared by the reaction of dilute H<sub>2</sub>SO<sub>4</sub> with metal sulphide.
- (a, b)
  - (b, c)
  - (c, d)
  - (a, c)

# Mcqs On Inorganic

**Yicheng Fang**



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