

Name: \_\_\_\_\_

Date: \_\_\_\_\_

## Student Exploration: Measuring Motion

**Vocabulary:** distance, speed

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

The **speed** of an animal is how fast it is moving. A speed of 5 m/s (meters per second) means that the animal moves a **distance** of 5 meters every second.

1. How would you measure the speed of an animal? \_\_\_\_\_  
\_\_\_\_\_
2. What do you think are the fastest animals? \_\_\_\_\_  
\_\_\_\_\_

### Gizmo Warm-up

You have been sent on an African safari by *International Geography* magazine. Your assignment is to find the fastest land animals in the world. Your only tool is a video camera.

The safari is shown in the *Measuring Motion* Gizmo™.

1. On the **Safari Adventure** tab, wait for an animal to pass by. Press the **record** button (📹). Press **stop** (⏏) when the animal has passed by.
  - A. Which animal did you record? \_\_\_\_\_
  - B. Was the animal walking or running? \_\_\_\_\_
2. Select the **Playback** tab. Practice using some of the different buttons on this tab:
  - Use the **Play** (▶) button to replay the simulation.
  - Click **Rewind** (⏮) to go back to the beginning.
  - Click **Advance frame** (⏭) to move forward exactly one second at a time.
  - Drag some **arrows** (📏) onto the recording to mark the positions of the animals at different times. (The arrows are on the left side of the Gizmo.)
  - Use the **Time slider** to go to a specific time.



# Measuring Motion Gizmo Answer Key

**Roman Wölfel**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

**Measuring Motion Gizmo Answer Key:**  
**Chicago Tribune Index ,2003**

## **Measuring Motion Gizmo Answer Key** Book Review: Unveiling the Power of Words

In some sort of driven by information and connectivity, the ability of words has become more evident than ever. They have the ability to inspire, provoke, and ignite change. Such may be the essence of the book **Measuring Motion Gizmo Answer Key**, a literary masterpiece that delves deep into the significance of words and their impact on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book's key themes, examine its writing style, and analyze its overall effect on readers.

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### **Table of Contents Measuring Motion Gizmo Answer Key**

1. Understanding the eBook Measuring Motion Gizmo Answer Key
  - The Rise of Digital Reading Measuring Motion Gizmo Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Measuring Motion Gizmo Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Measuring Motion Gizmo Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Measuring Motion Gizmo Answer Key
  - Personalized Recommendations
  - Measuring Motion Gizmo Answer Key User Reviews and Ratings
  - Measuring Motion Gizmo Answer Key and Bestseller Lists

5. Accessing Measuring Motion Gizmo Answer Key Free and Paid eBooks
  - Measuring Motion Gizmo Answer Key Public Domain eBooks
  - Measuring Motion Gizmo Answer Key eBook Subscription Services
  - Measuring Motion Gizmo Answer Key Budget-Friendly Options
6. Navigating Measuring Motion Gizmo Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Measuring Motion Gizmo Answer Key Compatibility with Devices
  - Measuring Motion Gizmo Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Measuring Motion Gizmo Answer Key
  - Highlighting and Note-Taking Measuring Motion Gizmo Answer Key
  - Interactive Elements Measuring Motion Gizmo Answer Key
8. Staying Engaged with Measuring Motion Gizmo Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Measuring Motion Gizmo Answer Key
9. Balancing eBooks and Physical Books Measuring Motion Gizmo Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Measuring Motion Gizmo Answer Key
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Measuring Motion Gizmo Answer Key
  - Setting Reading Goals Measuring Motion Gizmo Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Measuring Motion Gizmo Answer Key
  - Fact-Checking eBook Content of Measuring Motion Gizmo Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction

with these measurements bromobenzene was found to be the limiting reagent since only 4.56 ml of bromobenzene was used the amount of product predicted to be recovered was 8.7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction especially nitration of methyl benzoate to give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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then actual is your actual findings yield actual theoretical  $\times 100$  what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens

very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 °C

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1-bromo-4-nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture  $\text{HNO}_3$  3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accommodate their electrons demonstration of creating the p arenium cation by nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion  $\text{NO}_2^+$  which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4 0ml of nitric acid and 4 0ml of sulfuric acid and will then react with 4 5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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experimental spectra reaction a fluorobenzene stock fluorobenzene 1 h 1 h 19 f 19 f 13 c nmr fid for reference only sample 1 h 1 h 19 f 19 f 13 c nmr eas nitration of fluorobenzene crude product not available for submission for credit

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