Practicing the Mole - - Even Problems Answer Key

Calculate the mass in grams of each of the following:

2. 8.00 moles of aluminum

6. 7.00 moles of iodine (I2)

4. 2.00 x 10² moles of chlorine (Cl₂) 8. 9.20 moles of iron

Calculate the number of moles of atoms in each of the following:

10. 281 g of iron

14. 93.0 g of phosphorus

12. 72.0 g of magnesium

16. 1035 g of lead

Calculate the number of atoms in each of the following:

18. 5.40 moles of iron

20. 10.0 moles of titanium

Mixed Practice Mole Problems Worksheet Answer

Brendan G. Carr

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Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction The Effective Use of Analogies on Solving Mole Problems in High School Chemistry Virginia Di-Anna Sutula, 1988

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web oct 13 2023 how to run a successful church pledge campaign church pledge campaigns are a fantastic way to boost church giving overall often your congregants can give more over time than they can in a one time gift so collecting pledges helps encourage your donors to give deeply with a set deadline to complete their total gift