### SEPARATING MIXTURES

	Required materials	Describe what happens
	Method name: Density - Objects of different densities - Water	Place objects of different densities into water. Denser objects sink will hollow objects floot
0	Method name: Magnet  - Metal (Iron) - Nonmetal Items - Magnet	MIXTURES
	Method name Filter - Small particles - Liquid	Study Gu.
		Mixtures Aheten
	Method name Sieve	Solution Meture
		Disasta Breedy

## 11XTURES & SOLUTIONS

Study Guide

Rate each of the following skills: I - Very difficult to me Z - Need some help Y - I can teach a friend 3 - I can almost do this

I can distinguish between a mixture and solution. ...... I can describe materials that can dissolve in . I can explain how to separate mixtures and which method to use.

Word	Definition  A heterogeneous continuous of substances that keep their identity	
Mixtures		
Solution	Michare created when one substance desolves in another	
Dissolve	Breaks down and becomes part of a liquid	
Solvent	The liquid that dissolves the substance	

Highlight any phrones that are design time a solution.

- Includes only solid substances.
- Enchades Z bounds.
- Enchades a liquid and a nolid
- Most can be separated.
- Most cornel be separated
- Something descrives

Circula Sha minh man hades

# ANSWER KEY INCLUDED

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Small particles

Lorge porticles

#### to dissolve quicker?

- Increese the temperature of the **Britishing**
- Stip the substance in the **America**
- Cut the solute into smaler pieces- this creates more surface. dress.

#### Salt Sugar Coffee/cocoa

Name 3 substances that do NOT dissolve in worter.

> Sand Pepper



What can you expect to happen when you mix oil and water?

The oil may look like it dissolves at first, but it will separate from the worker and either sink or float.

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## **Mixtures And Solutions Study Guide**

**ML Morrison** 

#### **Mixtures And Solutions Study Guide:**

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter Student Study Guide/Solutions Manual for Essentials of General, Organic, and Biochemistry Denise Guinn, Rebecca Brewer, Rachel C. Lum, 2009-09-15 The Student Study Guide and Solutions Manual provides students with a combined manual designed to help them avoid common mistakes and understand key concepts After a brief review of each section's critical ideas students are taken through stepped out worked examples try it yourself examples and chapter guizzes all structured to reinforce chapter objectives and build problem solving techniques The solutions manual includes detailed solutions to all odd numbered exercises in the text Chemical Principles Student's Study Guide & Solutions Manual John Krenos, Joseph Potenza, Loretta Jones, Lynn Koplitz, Thomas Spence, 2004-03-19 This combination manual is designed to help students avoid common mistakes and understand the material better The solutions manual section includes detailed answers and explanations to the odd numbered exercises in the text Guide/Selected Solutions Manual Julie R. Frentrup, 2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book **Study Guide** and Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry, Third Edition John McMurry, Susan McMurry, 1999 Provides worked out solutions to text problems along with chapter by chapter outlines and a variety of self tests at the end of each chapter Basic Concepts of Chemistry, Study Guide Leo J. Malone, 2000 The text s three main goals are to introduce chemistry as a living relevant science to encourage learning and critical thinking and to help readers overcome the math difficulties that impede their progress in chemistry Designed to help readers master the principles of general chemistry As a prep book it promotes active involvement with the material There are special features throughout that reinforce concepts and help to develop strong problem solving and study skills Updated to Include an Interactive Learning Ware problems CD containing several of the chapter ending problems from the book in an interactive

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