

## Chemistry – Unit 5 Worksheet 1

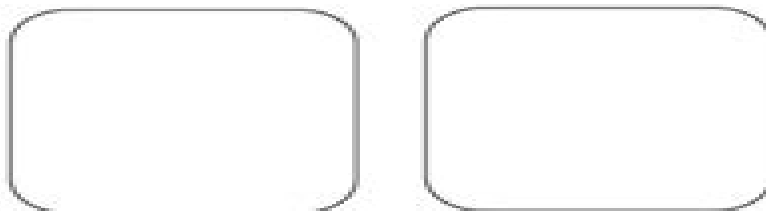
### Molar Masses of the Elements

#### Relative Mass From Gases

We have established that the combining ratio of gases can be explained if two assumptions are made:

1. Equal volumes of gases contain the same number of molecules at the same pressure and temperature.
2. Some pure elemental gases are clustered into pairs to form diatomic molecules.

Use these assumptions and particle diagrams to explain the fact that the density of oxygen gas at standard temperature and pressure is 1.43 g/liter, whereas the density of hydrogen gas under these conditions is 0.089 g/liter. How many times more massive is one molecule of oxygen than one molecule of hydrogen?



You shouldn't conclude that chemists were able to determine the molar masses of all the elements using this technique. Measurements of the density of the gaseous phase of many of the elements are difficult, if not impossible. However, we are going to see that chemists could use another tool – the *percent composition of compounds* to determine molar masses.

#### Relative Mass From Compounds

Many substances combine with oxygen to form a type of compound called an oxide. In the previous unit we saw that such combinations often occur in multiple proportions. John Dalton made the assumption that the lowest ratio was a 1:1 combination of elements. For now, we will make a similar assumption. *We may have to re-examine this assumption later.*

In Unit 4 you examined % composition data of several compounds. For example, 100 g of the simplest oxide of carbon contains 42.9g of C and 57.1 g of O. To be able to compare masses of elements to the mass of oxygen, it is important to use the same mass of oxygen in all of the ratios. Using a proportion, one can show that 75.0 g of C combines with 100 g of oxygen.

$$\frac{42.9\text{gC}}{57.1\text{gO}} = \frac{75.0\text{gC}}{100\text{gO}}$$

# Modeling Chemistry Unit 1 Objecives Matter

**A. Edward Ooghe**



## **Modeling Chemistry Unit 1 Objectives Matter:**

*Chemistry in the Community (ChemCom)* American Chemical Society, 2011-06-17 Touted as the most successful NSF funded project published *Chemistry in the Community ChemCom* by the American Chemical Society ACS offers a meaningful and memorable chemistry program for all levels of high school students ChemCom covers traditional chemistry topics within the context of societal issues and real world scenarios Centered on decision making activities where students are responsible for generating data in an investigating analyzing that data and then applying their chemistry knowledge to solve the presented problem The text is intensively laboratory based with all 39 of the investigations integrated within the text not separate from the reading With the ChemCom program students learn more organic and biochemistry more environmental and industrial chemistry and more on the particulate nature of matter than other textbooks all within the relevance of solving problems that arise in everyday life Meticulously updated to meet the needs of today's teachers and students the new sixth edition of ChemCom adheres to the new science framework as well as the forthcoming next generation of science standards Incorporating advances in learning and cognitive sciences ChemCom's wide ranging coverage builds upon the concepts and principles found in the National Science Education Standards Correlations are available showing how closely aligned ChemCom is to these and other state standards

**Handbook of CALL Teacher Education and Professional Development** Dara Tafazoli, Michelle Picard, 2023-04-26 This comprehensive handbook provides an overview of current trends in computer assisted language learning CALL teacher education and professional development across the globe It highlights theories and practices in CALL teacher education and professional development in five sections such as English language teaching including pre service teachers in service teachers teacher educators material developers course designers and researchers It explores the role of CALL teacher education and professional development in many underexplored countries such as Africa Asia Eastern Europe and the Middle East It stresses the critical role of professional development programs from the use of technology in its generic sense The theoretical and empirical chapters in the book provide a more inclusive and comprehensive picture of various aspects of CALL teacher education and professional development globally It offers context specific approaches and strategies to language teachers and teacher educators It provides pedagogical implications and suggestions for promoting digital literacy and autonomy in online education This book provides valuable insights for researchers teacher educators and teacher trainers in applied linguistics

**Chemistry** George Tyler Miller, 1987

**Cyber Science 6 Tm' 2007 Ed. ,** The World of Science Sixth Grade A. Edward Ooghe, 1963

**Resources in Education** , 1984 New Curriculum Perspectives for Junior High School Science Paul DeHart Hurd, 1970

**Research in Education** , 1973 **Prentice Hall Chemistry** , 2000 **Russian Journal of Physical Chemistry** , 2001 *Inquiry Into Physical Science* Willard J. Jacobson, 1969 Adult Training , 1975 Merrill Earth Science Ralph M. Feather, Susan Leach Snyder, Dale T. Hesser, 1995 Creating the Ideal School Albert Mamary, 2007-08-13

Creating the Ideal School is the only comprehensive guide available that provides all the tools necessary to create an ideal school. There are many education books out there but most are specific to one aspect of school improvement and do not provide a systemic framework or a total approach. This book does and it is based on expertise that was developed and implemented in a real school district with impressive results by author Albert Mamary. Under the author's leadership an under-achieving and troubled upstate New York school district was transformed into a nationally validated high performing school system and Dr Mamary was credited with creating the first and only comprehensive systemic framework for quality learning and school improvement. This book will be of interest to teachers and administrators who are interested in making profound improvements in education.

**The Chemistry of Galaxies** Marshall Lester McCall, 1982      **Encyclopedia of Soils in the Environment** Daniel Hillel, Jerry L. Hatfield, 2004. Provides a wide range of scientific knowledge on all aspects of soil science as well as the links of soils and soil science to environmental management, food production, biodiversity, climate change and many other areas of significant concern.

[Courses Catalog - University of Illinois at Urbana-Champaign](#)  
University of Illinois at Urbana-Champaign, 2002. Includes undergraduate and graduate courses.

[Teaching School Physics](#)  
John L. Lewis, 1972. A UNESCO source book.

**Radio Broadcast**, 1926      **The Electrical Journal**, 1909

## **Modeling Chemistry Unit 1 Objectives Matter** Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Modeling Chemistry Unit 1 Objectives Matter**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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