

Chemistry – Unit 7 Worksheet B

Write balanced chemical equations for the following reactions.

Hydrogen gas (collected in air) is also collected in (about) 1.5 times that amount (by volume).

1. Acetylene (C_2H_2) reacts with oxygen (hydrogen) + fluorine for three acetylenes, a fluorine.

$$3\text{C}_2\text{H}_2 + 5\text{F}_2 \rightarrow 6\text{HF} + 3\text{C}_2\text{F}_4$$

2. Calcium carbonate is decomposed (superheating) for three calcium oxide and carbon dioxide.

$$3\text{CaCO}_3 \rightarrow 3\text{CaO} + 3\text{CO}_2$$

3. Barium sulfate reacts with water for three barium hydroxide.

$$3\text{BaSO}_4 + 6\text{H}_2\text{O} \rightarrow 3\text{Ba(OH)}_2 + 3\text{H}_2\text{SO}_4$$

4. Methanethiol (CH_3SCH_3) decomposes to form methane (CH_4) and carbon monoxide.

$$2\text{CH}_3\text{SCH}_3 \rightarrow 2\text{CH}_4 + 2\text{CS}_2$$

5. Ethanol reacts with oxygen (O_2) to form ethyl acetate and water.

$$2\text{C}_2\text{H}_5\text{OH} + 2\text{O}_2 \rightarrow 2\text{C}_2\text{H}_5\text{O}_2 + 2\text{H}_2\text{O}$$

6. Calcium sulfate decomposes (superheating) for three calcium oxide and carbon dioxide.

$$3\text{CaSO}_4 \rightarrow 3\text{CaO} + 3\text{SO}_2$$

7. Iron reacts with sulfur gas (Fe_2S_3) to form iron (II) sulfide and hydrogen gas.

$$3\text{Fe} + 2\text{H}_2\text{S}_3 \rightarrow \text{Fe}_2\text{S}_3 + 3\text{H}_2$$

8. A nitrogen-containing carbon compound, $\text{C}_2\text{H}_5\text{N}_2$, decomposes to form ethane, C_2H_6 , and nitrogen gas.

$$2\text{C}_2\text{H}_5\text{N}_2 \rightarrow \text{C}_2\text{H}_6 + \text{N}_2$$

9. Phosphorus, P_4O_{10} , is formed when carbon monoxide reacts with phosphorus gas.

$$3\text{P}_4 + 10\text{CO} \rightarrow \text{P}_4\text{O}_{10} + 10\text{C}$$

Modeling Chemistry Unit 7 Ws3 Answers

Staley Wilbraham (Matta, Waterman)



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Pearson Chemistry Unit 7 Staley Wilbraham (Matta, Waterman),2022

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Table of Contents Modeling Chemistry Unit 7 Ws3 Answers

1. Understanding the eBook Modeling Chemistry Unit 7 Ws3 Answers
 - The Rise of Digital Reading Modeling Chemistry Unit 7 Ws3 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Modeling Chemistry Unit 7 Ws3 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Modeling Chemistry Unit 7 Ws3 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Modeling Chemistry Unit 7 Ws3 Answers
 - Personalized Recommendations
 - Modeling Chemistry Unit 7 Ws3 Answers User Reviews and Ratings
 - Modeling Chemistry Unit 7 Ws3 Answers and Bestseller Lists
5. Accessing Modeling Chemistry Unit 7 Ws3 Answers Free and Paid eBooks
 - Modeling Chemistry Unit 7 Ws3 Answers Public Domain eBooks
 - Modeling Chemistry Unit 7 Ws3 Answers eBook Subscription Services
 - Modeling Chemistry Unit 7 Ws3 Answers Budget-Friendly Options
6. Navigating Modeling Chemistry Unit 7 Ws3 Answers eBook Formats

- ePub, PDF, MOBI, and More
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 - Modeling Chemistry Unit 7 Ws3 Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Modeling Chemistry Unit 7 Ws3 Answers
 - Highlighting and Note-Taking Modeling Chemistry Unit 7 Ws3 Answers
 - Interactive Elements Modeling Chemistry Unit 7 Ws3 Answers
 8. Staying Engaged with Modeling Chemistry Unit 7 Ws3 Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Modeling Chemistry Unit 7 Ws3 Answers
 9. Balancing eBooks and Physical Books Modeling Chemistry Unit 7 Ws3 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Modeling Chemistry Unit 7 Ws3 Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Modeling Chemistry Unit 7 Ws3 Answers
 - Setting Reading Goals Modeling Chemistry Unit 7 Ws3 Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Modeling Chemistry Unit 7 Ws3 Answers
 - Fact-Checking eBook Content of Modeling Chemistry Unit 7 Ws3 Answers
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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