

Section 5.2 Review

The Structure of the Atom

Multiple Choice

For each question below, select the letter of the best answer.

- Which of the following statements about protons, neutrons, and electrons is *true*?
 - Protons and electrons have about the same mass.
 - Neutrons and electrons have equal but opposite charges.
 - The mass of a proton is nearly 2000 times greater than the mass of an electron.
 - The mass of an proton is nearly 2000 times greater than the mass of a neutron.
 - Neutrons and electrons are found in the nucleus of an atom.
- What is the atomic number of mercury, Hg?
 - 2+
 - 80
 - 120
 - 200
 - 200.6
- What is the mass number of an atom of chlorine with 18 neutrons?
 - 1-
 - 17
 - 18
 - 35
 - 35.5
- What element is represented by the Bohr-Rutherford diagram below?
 - potassium, K
 - calcium, Ca
 - hydrogen, H
 - argon, Ar
 - yttrium, Y
- Which of the following statements about the masses of atoms is *false*?
 - Most of the mass of an atom is located in the nucleus.
 - Electrons contribute very little to the mass of an atom.
 - Atoms of different elements have different masses.
 - Most of the mass of an atom's nucleus is due to the mass of neutrons.
 - The mass number of an atom is equal to the number of protons plus the number of neutrons.
- A neutral atom has eight electrons and eight neutrons. Which of the following options correctly represents the atom?
 - carbon-16
 - oxygen-16
 - oxygen-24
 - oxygen-8
 - sulfur-8
- Which of the following symbols represents the element with an atomic number of 14?
 - C
 - N
 - Ni
 - Si
 - S
- Which of the following statements is *false*?
 - The number of neutrons an atom has defines what element it is.
 - The number of electrons an atom has defines what element it is.
 - The number of protons an atom has defines what element it is.
 - All of the above are true.
 - One of the above are true.



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Yi-Tong Ma



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chapter 1 the solid state talks about two main categories of solids which are crystalline solids and amorphous solids before understanding the differences between these solids let us learn their characteristic properties they have a definite shape volume and mass

ncert solutions for class 12 chemistry chapter 1 the solid state - Jul 08 2022

there are three states of matter solid liquid and gas in chapter 1 of ncert solutions for class 12 chemistry solid characteristics are studied solids are defined by their mass shape and volume intermolecular distance and force are both short yet intermolecular force is strong