# MATTER—SUBSTANCES VS. MIXTURES

Tiron of Stollar

Name KEY

All matter can be classified as either a substance (element or compound) or a mixture (heterogeneous or homogeneous).

|   | Matte   | r                        |  |  |
|---|---|--------------------------|--|--|
| Substance<br>can write chemical<br>formula, homogeneous |   |                          | Mixtures<br>variable ratio                   |  |
| Element<br>one type<br>atom                             | Compound  two or more different atoms chemically bonded | Homogeneous<br>solutions | Heterogeneous<br>colloids and<br>suspensions |  |

Classify each of the following as to whether it is a substance or a mixture. If it is a substance, write Element or Compound in the substance column. If it is a mixture, write Heterogeneous or Homogeneous in the mixture column.

|      | Type of Matter     | Substance                       | Mixture                   |
|------|--------------------|---------------------------------|---------------------------|
| 1.   | chlorine           | Element . Ma compound           |                           |
| 2.   | water              | Compound                        |                           |
| 3.   | soll               |                                 | heterogeneou              |
| 4.   | sugar water        |                                 | homogeneous               |
| 5.   | oxygen             | Element or Compound             | ,                         |
| ó.   | carbon dioxide     | Element or Compound<br>Compound |                           |
| 7.   | rocky road loe cre |                                 | haterogeness              |
| 8.   | alcohol            | Compound                        |                           |
| 9.   | pure air           |                                 | homogeneous               |
| 10.  | iron               | Element.                        | /                         |
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J Rink

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