

**MATH SKILLS TRANSPARENCY WORKSHEET****11****Order of Operations****Use with Chapter 9,  
Section 9.1**

1. Follow the correct order of operations to solve these problems.

- a.  $4 + 2(9 + 1)$  \_\_\_\_\_  
b.  $9 \times 14 - (4 - 3)$  \_\_\_\_\_  
c.  $7(2 + 14/2) + 23$  \_\_\_\_\_  
d.  $12/4(2 - 1 + 4 \times 6)$  \_\_\_\_\_  
e.  $19 - 42/6 \times 2(3 + 4 - 3 \times 2)$  \_\_\_\_\_

2. Compare the answers to the following problems. Why is it important that parentheses be used?

a.  $16/4 + 4 = ?$

b.  $16/(4 + 4) = ?$

\_\_\_\_\_  
\_\_\_\_\_

3. For each of the following formulas, place parentheses where you think they should be. Explain your answers. If no parentheses are needed, write *NP*.

- a.  $\text{CaOH}_2$  \_\_\_\_\_  
b.  $\text{Na}_3\text{PO}_4$  \_\_\_\_\_  
c.  $\text{AlNO}_{33}$  \_\_\_\_\_  
d.  $\text{NH}_{42}\text{SO}_4$  \_\_\_\_\_  
e.  $\text{NiCO}_3$  \_\_\_\_\_

4. For each of the following formulas, list the number of each type of atom present and find the total number of atoms represented.

- a.  $\text{KBr}$  \_\_\_\_\_  
b.  $\text{Na}_2\text{O}$  \_\_\_\_\_  
c.  $\text{Au}(\text{OH})_3$  \_\_\_\_\_  
d.  $\text{NH}_4\text{NO}_3$  \_\_\_\_\_  
e.  $(\text{NH}_4)_3\text{PO}_4$  \_\_\_\_\_  
f.  $3\text{K}_2\text{SO}_4$  \_\_\_\_\_  
g.  $10\text{C}_6\text{H}_{12}\text{O}_6$  \_\_\_\_\_  
h.  $6\text{Mg}_2(\text{PO}_4)_2$  \_\_\_\_\_

# Math Skills Transparency Worksheet Answers Chapter 10

**Sterling Stuckey**



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