Moles and Chemical Formulas

11

A. Finding the Simplest Formula

- Mass of empty crucible + cover
- 2. Initial appearance of the magnesium
- 3. Mass of crucible + cover + magnesium
- 4. Mass of crucible + cover + oxide product

27, 842	g.
23 483	
23.073	- 8
29.3/7	_ g

Calculations

- 5. Mass of magnesium
- 6. Mass of magnesium compound
- 7. Mass of oxygen in the product
- 8. Moles of Mg (Show calculations.)

Moles of 0
(Show calculations.) 6.245 g x 1 wee = .3903 9. Moles of O

10. Which number of moles (Mg or O) is smaller _ U =

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Ralph J. Christensen

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