

৬। $A = x - 3$, $B = x^2 + 3x + 9$ এবং $C = x^3 - 27$ তিনটি বীজগণিতীয় রাশি।

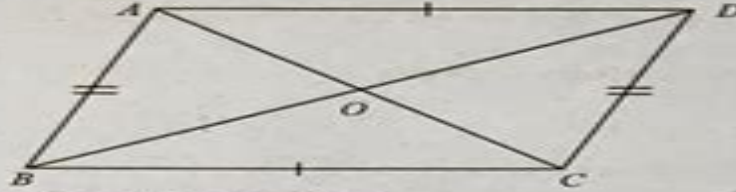
(ক) $\frac{x^3y^2 - x^2y^3}{x^3y - xy^3}$ কে লঘিষ্ঠ আকারে প্রকাশ কর।

(খ) প্রমাণ কর যে, $\frac{1}{A} \times \frac{x+3}{B} + \frac{x+3}{C} = 1$ ।

(গ) $\frac{1}{A} + \frac{1}{B} + \frac{1}{C}$ কে সাধারণ হরবিশিষ্ট ভগ্নাংশে প্রকাশ কর।

৭ বিভাগ—জ্যামিতি

৭।



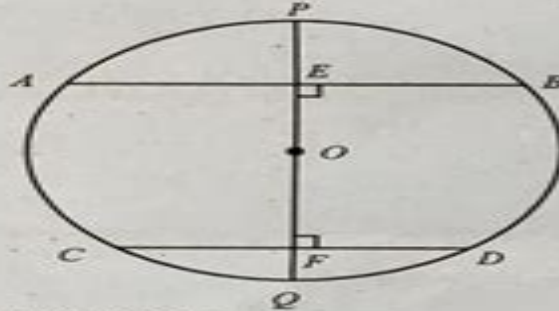
চিত্রে ABCD সামান্তরিকের AC ও BD দুইটি কর্ণ।

(ক) 4 সে.মি. দ্বারা বিশিষ্ট একটি ঘনকের সমগ্র পৃষ্ঠের ক্ষেত্রফল নির্ণয় কর।

(খ) প্রমাণ কর যে, $\angle ABC + \angle BCD + \angle ADC + \angle BAD =$ চার সমকোণ।

(গ) প্রমাণ কর যে, $AO = OC$ এবং $BO = OD$ ।

৮।



চিত্রে O বৃত্তের কেন্দ্র এবং PQ ব্যাস।

(ক) বৃত্তের ব্যাসার্ধ 4 সে.মি. হলে এর পরিধি নির্ণয় কর।

(খ) প্রমাণ কর যে, $PQ > CD$ ।

(গ) $AB > CD$ হলে, প্রমাণ কর যে, $OE < OF$ ।

৯। একটি চতুর্ভুজের দুটি সন্নিহিত বাহুর দৈর্ঘ্য $a = 5$ সে.মি., $b = 4$ সে.মি. এবং তিনটি কোণ $\angle x = 80^\circ$, $\angle y = 85^\circ$, $\angle z = 110^\circ$ ।

(ক) পেন্সিল ও কম্পাস ব্যবহার করে 45° কোণ অঙ্কন কর।

(খ) অঙ্কনের চিহ্ন ও বিবরণসহ চতুর্ভুজটি অঙ্কন কর।

(গ) এমন একটি রখাস অঙ্কন কর যার একটি বাহুর দৈর্ঘ্য a এর সমান এবং একটি কোণ $\angle x$ এর সমান।
[অঙ্কনের চিহ্ন ও বিবরণ আবশ্যিক]

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**John Richards,Manzoor Ahmed,Md.
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 $2 \text{ CO}_2(\text{g}) \rightarrow 2 \text{ C}(\text{s}) + 2 \text{ O}_2(\text{g}) \quad \Delta H = 94.5 \text{ kJ}$
 $\text{H}_2\text{O}(\text{g}) \rightarrow \text{H}_2(\text{g}) + \frac{1}{2} \text{ O}_2(\text{g}) \quad \Delta H = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) \rightarrow 2 \text{ C}(\text{s}) + 3 \text{ H}_2(\text{g}) \quad \Delta H = 283 \text{ kJ}$

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 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 2 \text{ H}_2\text{O}(\text{l}) \quad \Delta H = 1411 \text{ kJ/mole}$
 $\text{C}_2\text{H}_6(\text{g}) + 7/2 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{l}) \quad \Delta H = 1560 \text{ kJ/mole}$
 $\text{H}_2(\text{g}) + \frac{1}{2} \text{ O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l}) \quad \Delta H = 285 \text{ kJ/mole}$

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combustion of magnesium oxide Mg s $\frac{1}{2}\text{O}_2$ g MgO s we used

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