

Mole Ratios

How can the coefficients in a chemical equation be interpreted?

NATALIA
LACAYO
P. 2

Why?

A balanced chemical equation can tell us the number of reactant and product particles (ions, atoms, molecules or formula units) that are necessary to conserve mass during a chemical reaction. Typically when we balance the chemical equation we think in terms of individual particles. However, in real life the reaction represented by an equation occurs an unimaginable number of times. Short of writing very large numbers (10^{23} or larger) in front of each chemical in the equation, how can we interpret chemical equations so that they more realistically represent what is happening in real life? In this activity you will explore the different ways a chemical reaction can be interpreted.

Model 1 – A Chemical Reaction



1. Consider the reaction in Model 1.

a. What are the coefficients for each of the following substances in the reaction?

1 N_2

3 H_2

2 NH_3

b. Draw particle models below to illustrate the reaction in Model 1.



2. Consider each situation below as it relates to the reaction in Model 1.

a. Calculate the amount of reactants consumed and products made.

b. Record the ratio of N_2 to H_2 to NH_3 . Reduce the ratio to the lowest whole numbers possible.

	N_2 Consumed	H_2 Consumed	NH_3 Produced	Ratio $\text{N}_2:\text{H}_2:\text{NH}_3$ (reduced)
For a single reaction, how many molecules of each substance would be consumed or produced?	1	3	2	1:3:2
If the reaction occurred one hundred times, how many molecules would be consumed or produced?	100	300	200	1:3:2
If the reaction occurred 538 times, how many molecules would be consumed or produced?	538	1614	1076	1:3:2

Mole Ratios Model 1 A Chemical Reaction Answer Key

Michael Rajnik



Mole Ratios Model 1 A Chemical Reaction Answer Key:

Unveiling the Power of Verbal Artistry: An Psychological Sojourn through **Mole Ratios Model 1 A Chemical Reaction Answer Key**

In a global inundated with displays and the cacophony of immediate conversation, the profound power and emotional resonance of verbal beauty frequently diminish into obscurity, eclipsed by the regular assault of sound and distractions. Yet, located within the musical pages of **Mole Ratios Model 1 A Chemical Reaction Answer Key**, a captivating function of literary splendor that pulses with natural emotions, lies an memorable trip waiting to be embarked upon. Published by way of a virtuoso wordsmith, this magical opus guides visitors on a mental odyssey, lightly exposing the latent potential and profound influence embedded within the elaborate internet of language. Within the heart-wrenching expanse with this evocative examination, we will embark upon an introspective exploration of the book is main subjects, dissect its captivating publishing model, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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