

$$M_{\text{AgNO}_3} = \frac{n_{\text{AgNO}_3}}{V_{\text{AgNO}_3}} = \frac{25.513 \text{ g} + 149.87 \frac{\text{g}}{\text{mol}}}{100.00 \text{ mL}} + \frac{1000 \text{ mL}}{1 \text{ L}} = \boxed{1.569 \frac{\text{mol}}{\text{L}}}$$

4. Confirm that the molarity is the same molarity that you calculated in step 3.

Yes, the molarity is the same as the molarity calculated in step 3.

### Part B: Unknown solutions

Using a method similar to the one used in part A, create and carry out a procedure for determining the molarities of the unknown  $\text{AgNO}_3$  solution and the unknown  $\text{NaCl}$  solution. Round your data and show your calculations.

#### Unknown $\text{AgNO}_3$

Upon the addition of solid  $\text{NaCl}$ , solid  $\text{AgCl}$  was formed.

$\text{AgCl}$  formed (g) = 2.32019 g

$$\text{mass}(\text{AgNO}_3) = \frac{M_r(\text{AgNO}_3)}{M_r(\text{AgCl})} \times \text{mass}(\text{AgCl}) = \frac{149.87}{143.32} \frac{\text{g}}{\text{mol}} \times 2.32019 \text{ g} = \boxed{2.3500 \text{ g}}$$

$$M_{\text{AgNO}_3} = \frac{n_{\text{AgNO}_3}}{V_{\text{AgNO}_3}} = \frac{2.3500 \text{ g} + 149.87 \frac{\text{g}}{\text{mol}}}{100.00 \text{ mL}} + \frac{1000 \text{ mL}}{1 \text{ L}} = \boxed{0.16189 \frac{\text{mol}}{\text{L}}}$$

#### Unknown $\text{NaCl}$

Upon the addition of solid  $\text{AgNO}_3$ , solid  $\text{AgCl}$  was formed.

$\text{AgCl}$  formed (g) = 10.7907 g

$$\text{mass}(\text{NaCl}) = \frac{M_r(\text{NaCl})}{M_r(\text{AgCl})} \times \text{mass}(\text{AgCl}) = \frac{58.443}{143.32} \frac{\text{g}}{\text{mol}} \times 10.7907 \text{ g} = \boxed{4.4002 \text{ g}}$$

$$M_{\text{NaCl}} = \frac{n_{\text{NaCl}}}{V_{\text{NaCl}}} = \frac{4.4002 \text{ g} + 58.443 \frac{\text{g}}{\text{mol}}}{100.00 \text{ mL}} + \frac{1000 \text{ mL}}{1 \text{ L}} = \boxed{0.75294 \frac{\text{mol}}{\text{L}}}$$

### Conclusion questions

1. If you have a company that is using this method to create  $\text{AgCl}$ , knowing that  $\text{AgNO}_3$  is an expensive chemical, what do you think would be a better limiting reagent to use,  $\text{AgNO}_3$  or  $\text{NaCl}$ ? Why?

# Limited Reagent Lab Key

**DJ Losen**



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**Fundamentals of Chemical Reaction Engineering** Mark E. Davis, Robert J. Davis, 2013-05-27 Appropriate for a one semester undergraduate or first year graduate course this text introduces the quantitative treatment of chemical reaction engineering It covers both homogeneous and heterogeneous reacting systems and examines chemical reaction engineering as well as chemical reactor engineering Each chapter contains numerous worked out problems and real world vignettes involving commercial applications a feature widely praised by reviewers and teachers 2003 edition

**General Chemistry Key Concise Review Guide Sixth Edition** Darrell D. Ebbing, 1999 , **Chemistry in the Community (Enhanced Core Four)** American Chemical Society, 2006-02-15

*Strategic Applications of Named Reactions in Organic Synthesis* Laszlo Kurti, Barbara Czako, 2005-04-29 Kurti and Czako have produced an indispensable tool for specialists and non specialists in organic chemistry This innovative reference work includes 250 organic reactions and their strategic use in the synthesis of complex natural and unnatural products Reactions are thoroughly discussed in a convenient two page layout using full color Its comprehensive coverage superb organization quality of presentation and wealth of references make this a necessity for every organic chemist The first reference work on named reactions to present colored schemes for easier understanding 250 frequently used named reactions are presented in a convenient two page layout with numerous examples An opening list of abbreviations includes both structures and chemical names Contains more than 10 000 references grouped by seminal papers reviews modifications and theoretical works Appendices list reactions in order of discovery group by contemporary usage and provide additional study tools Extensive index quickly locates information using words found in text and drawings

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Chemical Reactor Design and Technology Hugo de Lasa,2012-12-06 Today s frustrations and anxieties resulting from two energy crises in only one decade show us the problems and fragility of a world built on high energy consumption accustomed to the use of cheap non renewable energy and to the acceptance of eXisting imbalances between the resources and demands of countries Despite all these stressing factors our world is still hesitatins about the urgency of undertaking new and decisive research that could stabilize our future Could this trend change in the near future In our view two different scenarios are possible A renewed energy tension could take place with an unpredictable timing mostly related to political and economic factors This could bring again scientists and technologists to a new state of shock and awaken our talents A second interesting and beneficial scenario could result from the positive influence of a new generation of researchers that with or without immediate crisis acting both in industry and academia will face the challenge of developing technologies and processes to pave the way to a less vulnerable society Because Chemical Reactor Design and Technology activities are at the heart of these required new technologies the timeliness of the NATO Advanced Study Institute at the University of Western Ontario London was very appropriate

**ChemCom** American Chemical Society,1993 *Semimicro Experiments in Organic Chemistry* Bruce Nelson Campbell,Ernest Ray Kline,1964 *SourceBook Version 2.1* ,1998 **Addison-Wesley Chemistry** ,1990 **Chemistry** Neil D. Jespersen,Alison Hyslop,2021-12-03 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the

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**Exploring Chemistry (Loose-Leaf)** Matthew Johll, 2012-05-04 Matthew Johll's Exploring Chemistry covers the standard topics for the nonmajors course in the typical order but each chapter unfolds in the context of a single case study that helps students connect what they are learning to real life situations For example students work through the often difficult topics of molecular structure gas laws and organic chemistry by learning about the development of powerful new chemotherapy drugs new technologies for screening airline passengers and the creation of biodegradable biopolymers It is the same case driven approach that Johll uses in his acclaimed Investigating Chemistry now in its Third Edition but Exploring Chemistry goes beyond the other book's specific focus on examples from forensic science to use real life stories from cooking athletics genetics green chemistry and more

**Wetland Design for Mining Operations** Thomas Wildeman, Gregory A. Brodie, James Gusek, 1993

**Essentials of Introductory Chemistry** Russo Steve Silver Michael, Steve Russo, 2001-12 Introductory Chemistry Third Edition helps readers master the quantitative skills and conceptual understanding they need to gain a deep understanding of chemistry Unlike other books on the market that emphasize rote memory of problem solving algorithms Introductory Chemistry takes a conceptual approach with the idea

that focusing on the concepts behind chemical equations helps readers become more proficient problem solvers What Is Chemistry The Numerical Side of Chemistry The Evolution of Atomic Theory The Modern Model of the Atom 1 Chemical Bonding and Nomenclature The Shape of Molecules Chemical Reactions Stoichiometry and the Mole The Transfer of Electrons from One Atom to Another in a Chemical Reaction Intermolecular Forces and the Phases of Matter What If There Were No Intermolecular Forces The Ideal Gas Solutions When Reactants Turn into Products Chemical Equilibrium Electrolytes Acids and Bases For all readers interested in introductory chemistry      **Advanced Control of Chemical Processes 1997 (ADCHEM'97)** Sirish L. Shah,Y. Arkun,1997 Paperback Advanced Control of Chemical Processes 1997 was an international event It attracted a total of 205 participants from industry and academia around the world Over 100 papers were presented at this symposium including 3 plenary addresses and 6 keynote talks The main themes included process monitoring pulp and paper process control model predictive control and modelling and simulation

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