

### Limiting & Excess Reactants

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Cheng, C., Hansen, T., McClelland, W., Salazar, M., Shapoval, P., & Warner, J. (2003). *Estuaries of Canada: A National Estuarine Research Network*.

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21. **Reaction mechanism:**  $\text{HBr} + \text{H}_2\text{O} \rightarrow \text{H}_3\text{O}^+ + \text{Br}^-$   
Detailed description: In the following reaction, the chemical equation is:  
$$\text{HBr} + \text{H}_2\text{O} \rightarrow \text{H}_3\text{O}^+ + \text{Br}^-$$
  
Describe the reaction mechanism using curly arrows to show electron flow. What does this tell you about the relative acidity of  $\text{HBr}$  compared to  $\text{H}_2\text{O}$ ? (4 marks)
22. **Reaction of an organic compound with sodium:**  $\text{C}_2\text{H}_5\text{Br} + 2\text{Na} \rightarrow \text{C}_2\text{H}_5\text{Na} + \text{NaBr}$   
Describe the reaction mechanism using curly arrows to show electron flow. What does this tell you about the relative acidity of  $\text{HBr}$  compared to  $\text{H}_2\text{O}$ ? (4 marks)
23. **Reaction of  $\text{C}_2\text{H}_5\text{Br}$  with  $\text{NaOH}$ :**  $\text{C}_2\text{H}_5\text{Br} + \text{NaOH} \rightarrow \text{C}_2\text{H}_5\text{OH} + \text{NaBr}$   
Describe the reaction mechanism using curly arrows to show electron flow. What does this tell you about the relative acidity of  $\text{HBr}$  compared to  $\text{H}_2\text{O}$ ? (4 marks)
24. **Reaction of  $\text{C}_2\text{H}_5\text{Br}$  with  $\text{NaOH}$  in  $\text{H}_2\text{O}$ :**  $\text{C}_2\text{H}_5\text{Br} + \text{NaOH} \rightarrow \text{C}_2\text{H}_5\text{OH} + \text{NaBr}$   
Describe the reaction mechanism using curly arrows to show electron flow. What does this tell you about the relative acidity of  $\text{HBr}$  compared to  $\text{H}_2\text{O}$ ? (4 marks)
25. **Reaction of  $\text{C}_2\text{H}_5\text{Br}$  with  $\text{NaOH}$  in  $\text{H}_2\text{O}$ :**  $\text{C}_2\text{H}_5\text{Br} + \text{NaOH} \rightarrow \text{C}_2\text{H}_5\text{OH} + \text{NaBr}$   
Describe the reaction mechanism using curly arrows to show electron flow. What does this tell you about the relative acidity of  $\text{HBr}$  compared to  $\text{H}_2\text{O}$ ? (4 marks)
26. **Reaction of  $\text{C}_2\text{H}_5\text{Br}$  with  $\text{NaOH}$  in  $\text{H}_2\text{O}$ :**  $\text{C}_2\text{H}_5\text{Br} + \text{NaOH} \rightarrow \text{C}_2\text{H}_5\text{OH} + \text{NaBr}$   
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Describe the reaction mechanism using curly arrows to show electron flow. What does this tell you about the relative acidity of  $\text{HBr}$  compared to  $\text{H}_2\text{O}$ ? (4 marks)
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Describe the reaction mechanism using curly arrows to show electron flow. What does this tell you about the relative acidity of  $\text{HBr}$  compared to  $\text{H}_2\text{O}$ ? (4 marks)
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30. **Reaction of  $\text{C}_2\text{H}_5\text{Br}$  with  $\text{NaOH}$  in  $\text{H}_2\text{O}$ :**  $\text{C}_2\text{H}_5\text{Br} + \text{NaOH} \rightarrow \text{C}_2\text{H}_5\text{OH} + \text{NaBr}$   
Describe the reaction mechanism using curly arrows to show electron flow. What does this tell you about the relative acidity of  $\text{HBr}$  compared to  $\text{H}_2\text{O}$ ? (4 marks)

# Limiting And Excess Reactants Packet Model 1

**Robert H. Perry, Don W. Green, James  
O. Maloney**



## **Limiting And Excess Reactants Packet Model 1:**

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Chemistry Stanley R. Radel, Marjorie H. Navidi, 1990      *Forensics in Chemistry* Sara McCubbins, Angela Codron, 2012 Forensics seems to have the unique ability to maintain student interest and promote content learning I still have students approach me from past years and ask about the forensics case and specific characters from the story I have never had a student come back to me and comment on that unit with the multiple choice test at the end from the Introduction to Forensics in Chemistry The Murder of Kirsten K How did Kirsten K's body wind up at the bottom of a lake and what do wedding cake ingredients soil samples radioactive decay bone age blood stains bullet matching and drug lab evidence reveal about whodunit These mysteries are at the core of this teacher resource book which meets the unique needs of high school chemistry classes in a highly memorable way The book makes forensic evidence the foundation of a series of eight hands on week long labs As you weave the labs throughout the year and students solve the case the narrative provides vivid lessons in why chemistry concepts are relevant and how they connect All chapters include case information specific to each performance assessment and highlight the related national standards and chemistry content Chapters provide Teacher guides to help you set up Student performance assessments A suspect file to introduce the characters and new information about their relationships to the case Samples of student work that has been previously assessed and that serves as an answer key for you Grading rubrics Using Forensics in Chemistry as your guide you will gain the confidence to use inquiry based strategies and performance based assessments with a complex chemistry curriculum Your students may gain an interest in chemistry that rivals their fascination with Bones and CSI      **Proceedings of the ... North American Thermal Analysis Society Conference** North American Thermal Analysis Society. Conference, 1984      **Lecture Outline-General Chemistry** Kenneth W. Whitten, Raymond E. Davis, Larry Peck, George G. Stanley, 2006-03 By Charles H Atwood and Kenneth W Whitten both of the University Georgia Richard M Hedges of Texas A M University and revised by Kimberly Schurmeier of the

University of Georgia Detailed lecture outlines of all of the text chapters are available to free students from tedious extensive note taking The outlines enable the students to listen more efficiently because they know that the important ideas and terms are already written down for them

**Water Distribution Modeling** Thomas M. Walski, Donald Vincent Chase, Dragan Savic, 2001 CD ROM contains WaterCAD software Exercise examination booklet

**Lecture Outl Gen Chem 7e** Davis, Peck, STANLE, 2003-07

*Prentice Hall Chemistry*, 2000

**The St. Louis Druggist**, 1883

**Industrial & Engineering Chemistry**, 1929

Pharmaceutical Record and Weekly Market Review P. W. Bedford, 1889

American Druggists' Circular and Chemical Gazette, 1921

**Russian Journal of General Chemistry**, 1997

**Physics Briefs**, 1994

The Chemical Trade Journal and Chemical Engineer, 1915

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Perry's Chemical Engineers' Handbook Robert H. Perry, Don W. Green, James O. Maloney, 1997

Reference work for chemical and process engineers Newest developments advances achievements and methods in various fields

*Scientific American*, 1877

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