Limiting Reactants and Percent Yield Worksheet

Name_____ Date_____ Period_____

CaCO₂ → CaO + CO₂

Calculate the theoretical yield of Calcium Oxide if 24.8 grams of Calcium Carbonate decomposes. Calculate the percent yield if you got 13.1 grams in the laboratory.

2) Al + CuSO₄ → Cu + Al₂(SO₄)₃

What is the percent yield if you experimentally produce 3.65 grams of copper when 1.87 grams of Aluminum reacts with 9.65 grams of Copper (II) Sulfate?

3) SO₂ + 2NaOH → Na₂SO₃ + H₂O

There are 100.0 grams of each reactant available. Determine the limiting reactant in this equation.

Limiting Reactant And Percent Yield Answer Key

R Barnett

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