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Lab Periodic Trends Worksheet

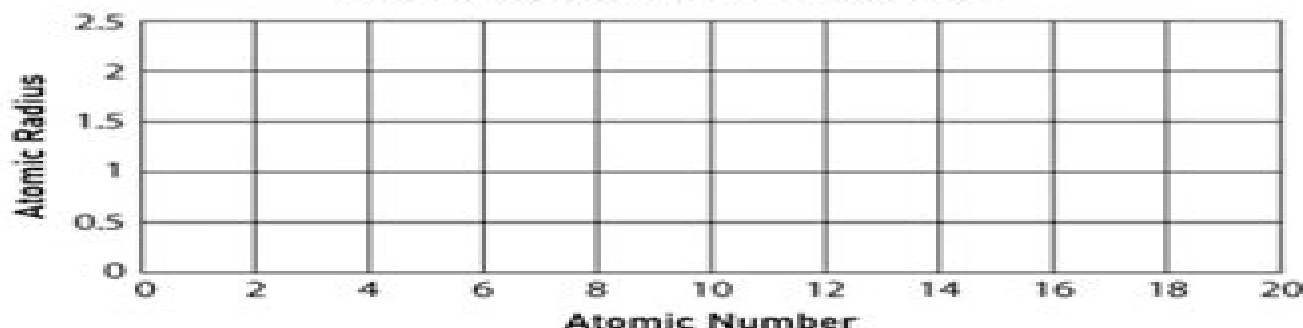
The atomic and the ionic radius of the first 18 elements of the periodic table are given below.

Element	H	He	Li	Be	B	C	N	O	F
Atomic Number	1	2	3	4	5	6	7	8	9
Atomic Radius (Å)	0.79	0.49	2.05	1.4	1.17	0.91	0.75	0.65	0.57
Ionic Radius (Å)	0.012	-	0.76	0.35	0.23	0.16	0.13	1.4	1.33

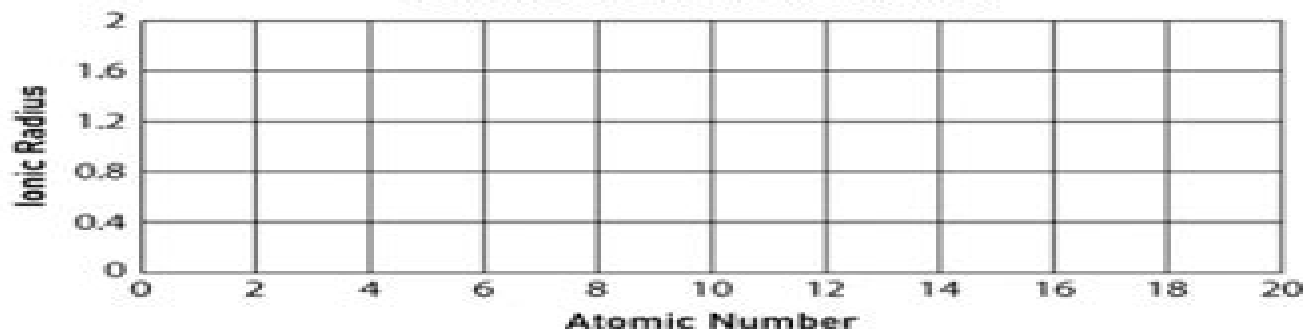
Ne	Na	Mg	Al	Si	P	S	Cl	Ar
10	11	12	13	14	15	16	17	18
0.38	2.23	1.72	1.82	1.46	1.23	1.09	0.97	0.71
-	1.02	0.72	0.535	0.4	0.38	0.37	1.81	-

Using the data above, construct the line graph of atomic radius vs. atomic number and ionic radius vs. the atomic number.

Atomic Radius vs. Atomic Number



Ionic Radius vs. Atomic Number



Lab Periodic Trends Key

KJ Lindholm-Leary

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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about the outcome of the experiment b To formulate a hypothesis c To think about a phenomenon in a specific way d To apply personal experience in answering a questions Pref **Prentice Hall Science Explorer** Michael J. Padilla,2002 **Merrill Chemistry** Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998 *The Software Encyclopedia 2001* ,2001

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