Limiting Reactant Lab

Pre-lab Questions: Use the background information in the lesson and the video to help you answer these questions before the lab.

1. What is a limiting reactant? (10pts)

A limiting reactant in a chemical reaction determines how much product will be produced.

2. What is an excess reactant? (10pts)

An excess reactant in a chemical reaction is present in an amount greater than what will be used to complete the reaction

What is the purpose of this experiment? (10pts)

to see the reaction of aluminum foil when placed in an aqueous solution of copper (II) chloride to produce copper and aluminum chloride. The purpose is to determine which of the reactants will limit the production of copper

Procedures: Use the video in the lesson to help you complete the data table and post-lab questions below.

Data & Obse	ervation Table (12 points)		
Mass of Al (g)	0.12 g		
Volume of CuCl ₂ (mL)	50.9 mL		
Color of CuCl ₂ solution before reaction	green		
Color of Al before reaction	Silver		
Color of CuCl ₂ solution after reaction	dark cloudy green		
Color of Al after the reaction	silver and brown		

4. Balance the equation below: (8 pts)

$$2$$
 Al(s) + 3 cucl₂(aq) \rightarrow 3 cu(s) + 2 AlCl₂(aq)

Limiting Reactant Lab Report Answers

John S. Phillips, Cheryl Wistrom

Limiting Reactant Lab Report Answers:

Laboratory Manual for Principles of General Chemistry Jo Allan Beran, 2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures Instructor's Guide for Introductory Chemistry in the Laboratory James F. Hall, 1996 Working with Chemistry Donald J. Wink, Sharon Fetzer-Gislason, Julie Ellefson Kuehn, 2004-02-20 With this modular laboratory program students build skills using important chemical concepts and techniques to the point where they are able to design a solution to a scenario drawn from a professional environment The scenarios are drawn from the lives of people who work with chemistry every day ranging from field ecologists to chemical engineers and include many health professionals as well

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