Limiting Reactant Worksheet

Balance the following reaction:

Suppose 100 grams of sodium chloride reacts with 200 g of barium phosphate. What is the limiting reactant?

[b] How many grams of excess reactant are left?

Acrylonitrile (C₃H₃N) is the starting material for the production of a kind of synthetic fiber acrylic. It can be made from propylene (C₃H₆) by reacting with nitric oxide (NO) as follows:

What mass of C₃H₃N can be made when 21.6 g of propylene react with 21.6 g of nitric oxide?

<u>Limiting Reagent And Percent Yield Worksheet Ib</u> <u>Chemistry</u>

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