Date	Name	
Section	Trans	
Instruction		

Pre-Lab Study Questions

1. Why are burning candles and rusting nails examples of chemical change?

Book to extentions

Contact man committee

They there go are of the theorem

Townstein in get because on of soder, decreation

of some transcent and, or that presented/outen seeds

2. What is included in a chemical equation?

a method and

3. How does a combination reaction differ from a decomposition reaction?

Company - -- elements contra to find company, becoming the -- the emperials spile

4. Balance each of the following reactions, and identify the type of reaction

Unbulanced Equation	Type of Reaction
a. ZAl(x) - Fe ₂ O ₁ (x) - 3 + Al ₂ O ₂ + Fe(i)	some massesses
B. (KClO ₁ (r) - 0 - KClO) + O ₂ (g)	decomposition.
$\epsilon_{-}/L_{-}(\epsilon) + Cl_{2}(g) \stackrel{h}{\longrightarrow} L_{1}Cl(\epsilon)$	dambinohan
d. $C_2H_4(g) + O_2(g) \xrightarrow{\Delta} CO_2(g) + H_2O(g)$	derle moderne
4. 3CrC+(ag) +2H2S(g) + Cr2S(g) + HCS(ag)	marke mydelesses

5. Complete and balance each of the following reactions

Reactants	Type of Reaction	
a. K(r) + N ₂ (g) ZKN(C)	Combination	
b. C ₃ H ₁₂ (g) + O ₂ (g) − ^h →	Combustion	
E. Al(s) + CuSO ₄ (ay)	Single Replacement	
d. CoCl ₂ (oq) - AgNO ₃ (oq)	Double Replacement	
e. MECO, (1) -1 - MO(6) + CO (1)	Decomposition	

Lab Questions For Chapter 5

Todd Lammle

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 - Chapter 2: Essential Elements of Lab Questions For Chapter 5
 - Chapter 3: Lab Questions For Chapter 5 in Everyday Life
 - Chapter 4: Lab Questions For Chapter 5 in Specific Contexts
 - ∘ Chapter 5: Conclusion
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