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Lab - Evidence for Chemical Change

One way of knowing that a chemical change has occurred is to observe that the properties of the products are different from those of the needants. The new product can their became a reaction; in another chemical reaction. In this expeniment you will observe a sequence of changes that occur when a solution that begins as copper (III) nitrate is treated with a series of different reactions. All of the reactions will take place in the same test tube. At each step you will look for evidence that a new substance is formed as a result of a chemical change. You will also observe heat changes and relate them to chemical reactions.

Provided:

- List four types of observations (evidence) that indicate when a chemical change has socured.
- For each reaction: IDENTIFY the type of chemical reaction, PRESECT the products, IDENTIFY the states of each substance, BALANCE the equation.

___ Cu()V()() () + ___ Ne()H() -- |

Cu()()H()() + | H()() -- |

Cu()() + ___ H()() -- |

Cu()() + ___ H()() -- |

Cu()() + ___ A(()) -- |

Resection Type:

____ HO(0+ ____ A(1)+

Reaction Type:

Lab Evidence For Chemical Change Answers

RS Peters

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Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science ,2003-11

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utilized and the education of the operators basic In fact most of the time these elements individually are perfectly legal to sell and possess However the combination of all these elements is what becomes the scene of illicit activity and a criminal operation In response to the increase in use of homemade explosive mixtures by terrorists both domestically and internationally the section clandestine manufacture of explosives is greatly enhanced Topics are presented in a manner which while detailed will not compromise the tactics techniques or procedures utilized by law enforcement and military personnel in their ability to combat the clandestine manufacture of contraband substances and the battle against domestic and international terrorism Key features Examines tell tale signs to look for in recognizing a clandestine lab Outlines how to safely process the site of a clandestine lab Details how to analyze collected evidence in the examination laboratory Provides quidelines as to what to derive from the physical evidence Offers specific tactics to effectively present the opinions associated with evidence that has been collected during the investigation in a written report military style briefing or to a jury in a legal proceeding Forensic Investigation of Clandestine Laboratories Second Edition guides the reader through the process of recognizing these illegal manufacturing operations. Then it examines the methods as to how to compile the volume of associated evidence into a package that can be presented in a court of law or to military commanders for decisive action It is an invaluable resource that will prove useful to chemistry lab technicians forensic investigators fire and first responder professionals military personnel police investigative agencies and narcotics units and lawyer trying cases involving clandestine labs Prentice Hall Science Explorer: Teacher's ed ,2005 **Prentice Hall Science Explorer** Michael J. Padilla,2002 Scientifica David Sang, Peter Ellis, Derek McMonagle, 2004 Bring your science lessons to life with Scientifica Providing just the right proportion of reading versus doing these engaging resources are differentiated to support and challenge pupils of varying abilities Educational Film & Video Locator of the Consortium of College and University Media Centers and R.R. Bowker Consortium of College and University Media Centers, 1990 CSE Report ,2006 Journal of Chemical Education ,1924 Includes Report of New England Association of Chemistry Teachers and Proceedings of the Pacific Southwest Association of Chemistry Teachers Chemical Interactions, 2005 Reports of Cases Decided in the Court of Appeals of the State of Georgia at the ... Georgia. Court of Appeals, 2001 Holt General Science William L. Ramsey, 1988 Holt Chemistry Ralph Thomas Myers, 2004 **Cooperative Learning in the** Chemistry Classroom Melissa Ann Flynn, 1999

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