4. (a. d):

4. (a, d)

NO, PbO₂

Generally transition elements and their salts are coloured due to the presence of unpaired electrons in metal ions. Which of the following compounds are coloured?

a. KMnO₄
b. Ce(SO₄)₂
c. TiCl₄
d. Cu₂Cl₂
Choose the correct option
1. (a, b)
2. (b, c)
3. (c, d)

General electronic configuration of actionoids is (n-2)f 1-14 (n-1)d⁰⁻²ns². Which of the following actinoids have one electron in 6d orbital?

a. U (Atomic no. 92)

b. Np (Atomic no. 93)

c. Pu (Atomic no. 94)

d. Am (Atomic no. 95)

Choose the correct option

1. (a, b)

2. (b, c)

3. (c, d)

Which of the following lanthanoids show +2 exidation state besides the characteristic exidation state +3 of lanthanoids?

(a) Ce
(b) Eu
(c) Yb
(d) Ho
Choose the correct option

1. (a, b)
2. (b, c)
3. (c, d)
4. (a, d)

On heating lead nitrate forms oxides of nitrogen and lead. The oxides formed are
1. N₂O, PbO

2. NO₂, PbO

3. NO. PbO

A brown ring is formed in the ring test for NO₃ ion. It is due to the formation of
[Fe (H₂O)₅ (NO)]²⁺

2. FeSO₄. NO₂

2. FeSO₄. NO₂ 3. [Fe (H₂O)₄ (NO)₂]²⁺ 4. FeSO₄. HNO₃

The oxidation state of the central atom in the anion of the compound NaH₂ PO₂ will be-1. +3 2. +5

3. +1 4. -3

A black compound of manganese reacts with a halogen acid to give greenish-yellow gas. When an excess of this gas reacts with NH₂ an unstable trihalide is formed. In this process, the oxidation state of nitrogen changes from-

1. -3 to +3 2. -3 to 0 3. -3 to +5 4. 0 to -3

3 Which of the following is isoelectronic pair?

1. ICl₂ , ClO₂ 2. BrO₂ , BrF₂ 3. ClO₂ , BrF 4. CN , O₃

9 Which of the following statements are correct for

SO₂ gas?

a. It acts as a bleaching agent in moist conditions.

Its molecule has linear geometry.

Its dilute solution is used as a disinfectant.

d. It can be prepared by the reaction of dilute H₂SO₄ with metal sulphide.

1. (a, b) 2. (b, c) 3. (c, d) 4. (a, c)

Mcqs On Inorganic

Julia Schneider



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