17.2 Measuring and Expressing Enthalpy Changes

Enthalpy Change in a Calorimetry Experiment

When 25.0 mL of water containing 0.025 mol HCI at 25.0°C is added to 25.0 mL of water containing 0.025 mol NaOH at 25.0°C in a foam-cup calorimeter, a reaction occurs. Calculate the enthalpy change (in kJ) during this reaction if the highest temperature observed is 32.0°C. Assume that the densities of the solutions are 1.00 g/mL and the volume of the final solution is equal to the sum of the volumes of the reacting solutions.

Measuring And Expressing Enthalpy Changes Answers

James M. Postma, Julian L. Roberts, J. Leland Hollenberg

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introduction to current knowledge techniques and applications The volume is essentially in two parts the first half covering a basic introduction to proteins appropriate at the undergraduate and early postgraduate level which will prove a valuable teaching aid The second half is a more advanced guide to concepts and methods covering a range of aspects not previously collected in one volume It will serve as a background reader and guide for advanced research study General Chemistry Raymond Chang,2003 Publisher Description Automatic Potentiometric Titrations G. Svehla,1978 International Series in Analytical Chemistry ,1975 General Chemistry Darrell D. Ebbing,Larry K. Krannich,Joan I. Senyk,1996

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Chemical and Physiological Abstracts ,1943 Starch and Its Derivatives Jack Augustus Radley,1968

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