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web we can write a balanced nuclear equation for this reaction as let s do a practice example identify the product x by writing a balanced nuclear equation ${}_{55}^{137}\text{Cs} \rightarrow {}_{56}^{137}\text{Ba} + \text{X}$ because the mass number is the same on both sides of the equation x must have a mass number of 0

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web set up and balance a nuclear reaction equation to find the mass number and atomic number of the second nuclide remember that the two ${}_{0}^{1}\text{n}$ contribute

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web in balancing a nuclear equation it is important to remember that the sum of all the mass numbers and atomic numbers given on the upper left and lower left side of the element symbol respectively must be equal for both sides of the equation

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web nuclear reactions are going on all around us using correctly balanced equations is important when trying to understand nuclear reactions all equations need to be balanced to conform to two conservation laws the mass number is conserved and the electrical charge of the nucleus is conserved