# Homework 1: Mixed Ionic/Covalent Compound Naming

For each of the following questions, determine whether the compound is ionic or covalent and name it appropriately.

- Na<sub>2</sub>CO<sub>3</sub>
- P<sub>2</sub>O<sub>5</sub>
- NH<sub>3</sub>
- FeSO<sub>4</sub>
- 5iO<sub>2</sub>
- 6) GaCl<sub>3</sub>
- 7) CoBr<sub>2</sub>
- 8) B<sub>2</sub>H<sub>4</sub>
- 9) CO
- 10) P4

For each of the following questions, determine whether the compound is ionic or covalent and write the appropriate formula for it.

- dinitrogen trioxide
- 12) nitrogen
- 13) methane
- 14) lithium acetate
- 15) phosphorus trifluoride
- 16) vanadium (V) oxide
- 17) aluminum hydroxide
- zinc sulfide
- 19) silicon tetrafluoride
- silver phosphate

# **Mixed Ionic Compound Homework**

**AW Rasmussen** 



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