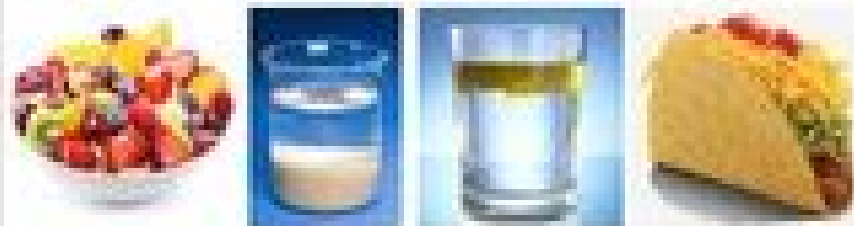


# Mixtures & Solutions

## MIXTURE

- A combination of two or more substances that keep their identities.
- The parts of a mixture don't undergo a chemical change.
- Can see the individual parts.

**Examples:** fruit salad, taco, sand and water, oil and water.



## SOLUTION

- A mixture that has the same composition throughout.
- Forms when one ingredient dissolves in another and particles are evenly mixed.
- Cannot see dissolved particles.

**Examples:** sugar and water, lemonade, Koolaid, salt water.



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**Journal of Comparative Pathology and Therapeutics**, 1928 *A Dictionary of Applied Chemistry* Thomas Edward Thorpe, 1894 Journal of the Chemical Society, 1913 Modern Chemistry Nicholas D. Tzimopoulos, 1993 **Journal of the Society of Chemical Industry** Society of Chemical Industry (Great Britain), 1919 **The Journal of the Society of Chemical Industry**, 1906 **Scientific American**, 1890 **Objective NCERT Xtract Chemistry for NEET/ JEE Main 5th Edition** Disha Experts, *Facts about Sugar*, 1924 The Principles of chemistry pt. 2 Dmitry Ivanovich Mendeleev, 1902

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