

# CALIBRATION MANUAL



AVERY BERKEL L122 HL122



# Manual Setting Avery Berkel Hl 122

**K Morrison**



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microscopes you can simply use your fingers purpose determine the average atomic mass for the fictitious element beanium  
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beans mass of beans of this kind atomic mass of this bean isotope abundance kidney bean 32 36 1 122 black eye bean 98 23  
0 25 white bean 30 16 0 50 calculation of atomic mass of beanium 36 32 1 amu

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atoms total mass isotope 1 atoms 1 avg mass 1 total mass isotope 2 atoms 2 avg mass 2 total mass isotope 3 atoms 3 avg  
mass 3 total mass of all atoms total of atoms

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web isotopes and weighted average atomic mass amu atomic mass unit mass of one proton mass of one neutron a sample of  
cesium cs has the following abundance cs 132 20 0 cs 133 75 3 cs 134 4 7 weighted average atomic mass  $132 \times 0.200$   $133 \times 0.752$   $134 \times 0.047$  132 85 amu

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element are isotopes of each other ordinary beans are a lot bigger than atoms but perhaps they can give you one or two clues  
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all beans 100 beanium isotope blackium beanium isotope brownium beanium

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web beanium isotope b brownium beanium isotope c whiteium of atoms beans mass beans in cup mass empty cup mass of  
atoms beans 2 minus 3 average mass of 1 bean 4 divided by 1 of beans 1 a b c  $\times 100$  section 3 calculate the average atomic  
mass of beanium

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simulation form you will be asked to gather data about the isotopes and organize the data if atoms were as large as beans  
they could be sorted counted and massed

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web 1 obtain a sample of beanium separate the pintonium navium and lentilium and then mass each group record the values in the data table 2 count the number of each isotope by counting the pintonium navium and lentilium 3 calculate the total number of isotopes

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web procedure obtain a sample of the classroom mixture of beanium sort your beanium into its three isotopic bean types count the number of beans in each pile find the mass of each pile of beans determine the average mass of each type of bean based on the samples masses

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web terms in this set 14 what are the objective of the lab calculate from experimental data the atomic mass of the fictitious element beanium by using a weighted average and simple average decide which method gives a better representation of the atomic mass of beanium

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web good luck saftey 1 do not eat the beans 2 do not throw the beans 3 do not crush the beans don t mess with the beans pre lab questions 1 there are many beans in your container each bean represents an 2 there are three kinds of beans in your container each kind represents a different

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