

## BUILD Vocabulary

**ionic bond** chemical bond formed when one or more electrons are transferred from one atom to another

**ion** atom that has a positive or negative charge

**covalent bond** type of bond between atoms in which the electrons are shared

**molecule** smallest unit of most compounds that displays all the properties of that compound

**van der Waals forces** weak attraction that develops between oppositely charged regions of nearby molecules

**Preface** The prefix *co-* and *co-* mean “together.” A compound is made up of elements joined together, and a covalent bond connects atoms that share electrons together. **How are covalent bonds different from ionic bonds?**

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## Chemical Bonds

**KEY QUESTION** What are the main types of chemical bonds?

The atoms in compounds are held together by chemical bonds. Bonds are formed by the electrons in the outer shell, which are called valence electrons. The main types of chemical bonds are ionic bonds and covalent bonds.

**Ionic Bonds** **Ionic bonds** are formed when one or more electrons are transferred from one atom to another. This transfer causes the atom losing the electrons to become positively charged. The atom gaining the electrons becomes negatively charged. The positively and negatively charged atoms become **ions**. The oppositely charged ions are attracted to each other and form an ionic bond.

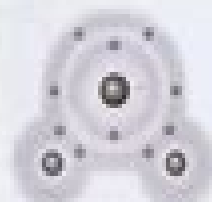
**Covalent Bonds** Sometimes atoms share electrons rather than transferring them. These shared electrons move around the nuclei of both atoms. The bond formed by sharing electrons is called a **covalent bond**. Atoms can share between two and six electrons. The structure that results is known as a **molecule**. A molecule is the smallest unit of most compounds.

**Weak Interactions** The strongest chemical bonds are ionic or covalent. Weak interactions between atoms are also important. Within living organisms, many molecules interact only briefly. **Van der Waals forces** are one type of weak interaction between molecules. They occur when molecules are very close together.

Hydrogen bonds are another type of weak interaction. They occur between a hydrogen atom of one molecule and an oxygen or nitrogen atom of a neighboring molecule.

### Visual Reading Skill Covalent Bonds

1. What kind of chemical bond holds the water molecule together? \_\_\_\_\_
2. What particles are shared between the atoms in the molecule? \_\_\_\_\_
3. What is the other main type of chemical bond? \_\_\_\_\_
4. What is the difference between covalent and ionic bonds? \_\_\_\_\_



Water molecule ( $H_2O$ )

## Properties of Water

**READING TOOL** Cause and Effect As you read the part of the lesson under “The Water Molecule,” use the table below to list the causes and effects of the properties of water. Many of the causes may be the same, but each characteristic will have a different effect. Part of the table is completed for you.

Characteristic of Water	Cause	Effect
Hydrogen Bonding		The slightly negative end of a water molecule is attracted to the slightly positive end of another water molecule.
Expands Upon Freezing		
Dissolves Many Substances	Water forms hydrogen bonds.	
Cohesion		
Adhesion		
Heat Capacity		

# Miller And Levine Biology Workbook Answers Chapter 16

**L Towne**



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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Miller And Levine Biology Workbook Answers Chapter 16**," an enthralling opus penned by a very acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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web sep 24 2023 this module teaches you how to develop or improve device drivers in the linux kernel for projects on embedded platforms or on the traditional pc platform in no time through theory and practical the module makes you familiar with the essentials of kernel development kernel architecture the main apis integration of device drivers