

Mixed Nomenclature

Identify each compound as Ionic, Covalent, or Acid, then complete the missing information.

	Type of Compound (Ionic, Covalent, or Acid)	Name	Chemical Formula
1.	Ionic	⁺² Magnesium sulfide ⁻²	MgS
2.	Covalent	Tetrasulfur octaoxide	S ₄ O ₈
3.	Ionic	Magnesium chloride	MgCl ₂
4.	Ionic	⁺¹ Cesium carbonate ⁻²	Cs ₂ CO ₃
5.	Acid	Hydrofluoric acid	HF
6.	Covalent	Octanitrogen hexaoxide	N ₈ O ₆
7.	Ionic	⁺² Manganese (II) chloride ⁻¹	MnCl ₂
8.	Acid	Phosphoric acid	H ₃ PO ₄
9.	Ionic	⁺¹ Ammonium sulfate ⁻²	(NH ₄) ₂ SO ₄
10.	Ionic	Iron (III) sulfide	Fe ₂ S ₃
11.	Acid	Chloric acid	HClO ₃
12.	Ionic	⁺³ Aluminum chlorate ⁻¹	Al(ClO ₃) ₃
13.	Covalent	Dinitrogen trioxide	N ₂ O ₃
14.	Ionic	⁺¹ Ammonium nitrite ⁻¹	NH ₄ NO ₂
15.	Ionic	Sodium nitride	Na ₃ N
16.	Covalent	Trisulfur nonafluoride	S ₃ F ₉
17.	Ionic	Copper (II) acetate	Cu(C ₂ H ₃ O ₂) ₂
18.	Ionic	Calcium phosphate	Ca ₃ (PO ₄) ₂
19.	Acid	Nitrous acid	HNO ₂
20.	Ionic	⁺² Beryllium cyanide ⁻¹	Be(CN) ₂
21.	Covalent	Phosphorus pentabromide	PBr ₅
22.	Ionic	⁺⁴ Tin (IV) carbonate ⁻²	Sn(CO ₃) ₂
23.	Acid	Hydronitric acid	H ₃ N
24.	Ionic	Nickel (II) sulfide	NiS
25.	Covalent	Tetrachlorine heptabromide	Cl ₄ Br ₇

Mixed Nomenclature And Properties Worksheet

IM Harris



Mixed Nomenclature And Properties Worksheet:

Mixed Nomenclature And Properties Worksheet Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the energy of words has been much more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such may be the essence of the book **Mixed Nomenclature And Properties Worksheet**, a literary masterpiece that delves deep into the significance of words and their impact on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book's key themes, examine its writing style, and analyze its overall effect on readers.

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