Name:	Date:
Chapter 5 Quiz Biol	ogy
The ability to do work	A. Activation energy (enzymes)
2. Stored energy that can be used for motion	B. Enzymes
3. Energy of motion	C. Catabolism
4. Energy conversion to heat, Heat energy	D. Anabolism
5. The study of energy or heat	E. Potential energy
6. Can never be created nor destroyed, amount of energy, in the universe remains constant	F. 1st law of thermodynamics
<ol> <li>Amount of energy transforming from potential to heat energy, or entropy in universe is increasing</li> </ol>	G. Reducing power
8. The starting molecules of a chemical reactions	H. Reactants and Substrates
9. Molecules at the end of a chemical reaction	I. Oxidation
<ol> <li>Reactions have products w/ more energy than the reactants (uses &amp; stores energy)</li> <li>spontaneous</li> </ol>	J. Products
11. An initial input of energy by destabilizing existing chemical bonds	K. B/w phosphate bonds
12. reactions become more spontaneous if their activation is lowered	L. Thermodynamics
13. Catalysts used by cells to perform particular reactions	M. Endergonic
14. Enzymes lower activation energy by binding specific molecules and stressing the bonds to make the reaction more likely to proceed	N. Coupled reactions
15. includes all the chemical and physical process continuously occurring in living cells	O. Biochemical Pathways
<ol> <li>Is the process by which food is changed into living tissue</li> </ol>	P. Redox Reduction
17. Is the process by which food is converted into energy and wastes of a simpler chemical composition	Q. Catalysis
18. Series of reactions of the product of one reaction is the substrate of the next	R. Release energy
19. What do free "inorganic" phosphates do?	S. Activation energy
20. When an atom or molecule loses an electron	T. Energy
21. When an atom or molecule gain an electron	U. How energy is measured
22. Have "the power" causes something else to be reduced	V. Kinetic energy
23. Where is energy "Stored" in ATP molecule?	W. Reduction
24. electrons pass from atoms or molecules to one another as part of many energy reactions	X. 2nd law of thermodynamics
25 Provide energy for the initiation of endergonic reactions	Y Metabolism

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**Jeffrey C. Pommerville** 

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